Pyriproxyfen: 95737-68-1 January 2012
Doc IIIA RMS: NL

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European Commission



Pyriproxyfen

CAS number 95737-68-1

Document III-A

Section 1 & 2 : Applicant & Identity
Study Summaries
Active Substance

Rapporteur Member State: The Netherlands January 2012

Draft CA-report and Proposed Decision of The Netherlands in the context of the Possible inclusion of pyriproxyfen in Annex I of Council Directive 98/8/EC

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Please refer to "Technical Notes for Guidance on Dossier Preparation including preparation and evaluation of study summaries under Directive 98/8 EC Concerning the Placing of Biocidal Products on the Market (Appendix 7.1 and 7.2)" for a list of the Standard Terms and Abbreviations used in this document.

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Section A1

1. Applicant

Annex Point IIA1

1.1 Applicant

Sumitomo Chemical (UK) PLC

Horatio House 75-85 Fulham Palace Road London W6 8JA United Kingdom

Telephone: +44 (0)208 600 7713

1.2 Manufacturer of Active Substance (if different)

Sumitomo Chemical Co. Ltd.



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Section A2 2. Identity of Active Substance

Subsection (Annex Point)		Official use only
2.1 Common name (IIA2.1)	Pyriproxyfen	
2.2 Chemical name (IIA2.2)	IUPAC: 4-phenoxyphenyl (RS)-2-(2-pyridyloxy)propyl ether CA: 4-phenoxyphenyl (RS)-2-(2-pyridyloxy)propyl ether	X1
2.3 Manufacturer's development code number(s) (IIA2.3)		
2.4 CAS No and EC numbers (IIA2.4)	Non-entry field	
2.4.1 CAS-No	95737-68-1	
2.4.2 EC-No 2.4.3 Other	429-800-1 (ELINCS) 715 (CIPAC)	

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25 Molecular and structural formula, molecular mass (IIA25)

Non-entry field

25.1 Molecular formula

C20H19NO3

25.2 Structural formula

$$\begin{array}{c|c} & & & \\ &$$

25.3 Molecular mass 321.37

2.6 Method of manufacture of the active substance (IIA21)

Refer to confidential information presented in the Confidential

Appendix

2.7 Specification of the purity of the active substance, as appropriate (IIA2.7)

Refer to confidential information presented in the Confidential

Appendix

2.8 Identity of impurities and additives, as appropriate (IIA2.8)

Refer to confidential information presented in the Confidential

Appendix

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2.9 The origin of the natural active substance or the precursor(s) of the active substance (IIA2.9)	Not relevant, as the active substance is not naturally occurring	
2.10	Exposure data in conformity with Annex VIIA to Council Directive 92/32/EEC (OJ No L, 05.06.1992, p. 1) amending Council Directive 67/548/EEC	
2.10.1 Human exposure towards active substance	Not relevant as active substance as manufactured is not produced in the EU	
2.10.1.1 Production		
i) Description of process	Not relevant as active substance as manufactured is not produced in the EU	
ii) Workplace description	Not relevant as active substance as manufactured is not produced in the EU	
iii) Inhalation exposure	Not relevant as active substance as manufactured is not produced in the EU	
iv) Dermal exposure	Not relevant as active substance as manufactured is not produced in the EU	
210121 ()		
1. Professional		
1. Professional Users i) Description of application process	For detailed description of intended uses refer to IIIB, Section 5	
1. Professional Users i) Description of	For detailed description of intended uses refer to IIIB, Section 5 For detailed description of intended uses refer to IIIB, Section 5	
1. Professional Users i) Description of application process ii) Workplace description iii) Inhalation	•	
Users i) Description of application process ii) Workplace description	For detailed description of intended uses refer to IIIB, Section 5	
1. Professional Users i) Description of application process ii) Workplace description iii) Inhalation exposure iv) Dermal	For detailed description of intended uses refer to IIIB, Section 5 For evaluation of primary (direct) exposure refer to IIB, Section 8.2	
1. Professional Users i) Description of application process ii) Workplace description iii) Inhalation exposure iv) Dermal exposure 2. Non-professional Users including the general public (i) via inhalational	For detailed description of intended uses refer to IIIB, Section 5 For evaluation of primary (direct) exposure refer to IIB, Section 8.2	
1. Professional Users i) Description of application process ii) Workplace description iii) Inhalation exposure iv) Dermal exposure 2. Non-professional Users including the general public	For detailed description of intended uses refer to IIIB, Section 5 For evaluation of primary (direct) exposure refer to IIB, Section 8.2 For evaluation of primary (direct) exposure refer to IIB, Section 8.2	
1. Professional Users i) Description of application process ii) Workplace description iii) Inhalation exposure iv) Dermal exposure 2. Non-professional Users including the general public (i) via inhalational contact	For detailed description of intended uses refer to IIIB, Section 5 For evaluation of primary (direct) exposure refer to IIB, Section 8.2 For evaluation of primary (direct) exposure refer to IIB, Section 8.2 For evaluation of primary (direct) exposure refer to IIB, Section 8.2	

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(v) indirect via

environment For evaluation of secondary (indirect) exposure refer to IIB, Section

8.2

2.10.2 Environmental exposure towards active substance

2.10.2.1 Production

(i) Releases into water

Not relevant as active substance as manufactured is not produced in

the EU

(ii) Releases into air

Not relevant as active substance as manufactured is not produced in

the EU

(iii) Waste disposal

Not relevant as active substance as manufactured is not produced in

the EU

2.10.2.2 Intended use(s)

Affected compartment(s): For evaluation of exposure to environmental compartments refer to

IIB, Section 8.3

Water For evaluation of exposure to environmental compartments refer to

IIB, Section 8.3

Sediment For evaluation of exposure to environmental compartments refer to

IIB, Section 8.3

Air For evaluation of exposure to environmental compartments refer to

IIB, Section 8.3

Soil For evaluation of exposure to environmental compartments refer to

IIB, Section 8.3

Predicted concentration in

the affected compartment(s)

For evaluation of exposure to environmental compartments refer to

IIB, Section 8.3

Water For evaluation of exposure to environmental compartments refer to

IIB, Section 8.3

Sediment For evaluation of exposure to environmental compartments refer to

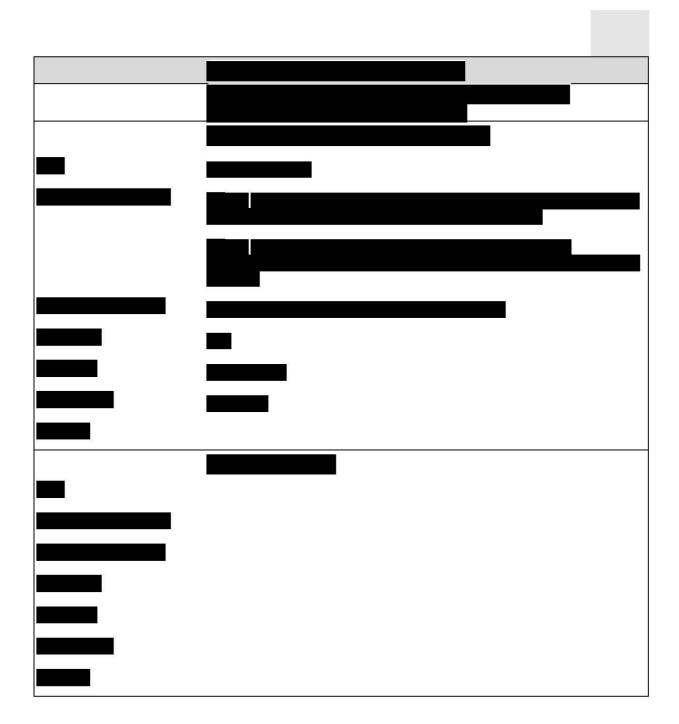
IIB, Section 8.3

Air For evaluation of exposure to environmental compartments refer to

IIB, Section 8.3

Soil For evaluation of exposure to environmental compartments refer to

IIB, Section 8.3



European Commission



Pyriproxyfen

CAS number 95737-68-1

Document III-A

Section 3 Phys-Chem Study Summaries Active Substance

Rapporteur Member State: The Netherlands January 2012

Draft CA-report and Proposed Decision of The Netherlands in the context of the Possible inclusion of pyriproxyfen in Annex I of Council Directive 98/8/EC

Pyriproxyfen: CAS number 95737-68-1	January 2012
Doc.IIIA - Section 3 Phys-Chem - Active substance	RMS: NL

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Please refer to "Technical Notes for Guidance on Dossier Preparation including preparation and evaluation of study summaries under Directive 98/8 EC Concerning the Placing of Biocidal Products on the Market (Appendix 7.1 and 7.2)" for a list of the Standard Terms and Abbreviations used in this document.

Section A3 Physical and Chemical Properties of Active Substance

BUU	on A3	rhysical and Chemical Properties of Active Substance								
	Subsection (Annex Point)	Method	Purity/ Specification	Results	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only	
3.1	Melting point, boiling point, relative density (IIA3.1)	-	-	-	-	-	-	-		
3.1.1	Melting point	OECD 102 equivalent to EEC A.1 - Heated metal block with capillary tube		Melting point: 48.0-50.0°C	-					
3.1.2	Boiling point	OECD 103 equivalent to EEC A.2 - Differential thermal analysis		Boiling point: 318°C	-					
3.1.3	Bulk density/ relative density	Gas comparison pyknometer method - equivalent to EEC A.3 OECD 109		Density: 1.26 g/cm ³ at 23°C	-					
		Gas comparison pyknometer method - equivalent to EEC A.3 OECD 109		Relative density 1.143	-		,			
3.2	Vapour pressure (IIA3.2)	OECD104 equivalent to EEC A.4 - Gas saturation method		Vapour pressure: <1.33 x 10 ⁻⁵ Pa (<1.0 x 10 ⁻⁷ mmHg) at 22.81°C	=					

Section A3 Physical and Chemical Properties of Active Substance

~ ~ ~ ~ ~	on A5	Thysical and Chemical Troperties of Active Substance								
	Subsection (Annex Point)	Method	Purity/ Specification	Results	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only	
3.2.1	Henry's Law Constant (Pt. I-A3.2)	Calculation	-	Henry's law constant (calculated): <1.17 x 10 ⁻² Pa m ³ mol ⁻¹ (<1.16 x 10 ⁻⁷ atm mol ⁻¹ m ³) at 22-25°C				F		
3.3	Appearance (IIA3.3)	-		-	-	-	:= :::::::::::::::::::::::::::::::::::	1-6		
3.3.1	Physical state	Physical state: In-house method (Visual inspection)*		Granular solid at 25°C			•			
		Physical state: In-house method (Visual inspection)*		Solid at 20°C						
3.3.2	Colour	Colour: ASTM D1535-80*		Colour: White (Munsell colour: N9.5/90.0%R)	-					

Section A3 Physical and Chemical Properties of Active Substance

5000	ion A5	rnystem und enem	iteal Froperities of Active Substance						
	Subsection (Annex Point)	Method	Purity/ Specification	Results	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
		Colour: ASTM D1535-68*		Colour: Pale yellowish white (Munsell colour: 5Y(9/2))	-				
3.3.3	Odour	ASTM D1292-86*		Odourless	-	•			
		General Notices of the Pharmacopoeia of Japan, 11 th edition*		Faint characteristic odour	-				
3.4	Absorption spectra (IIA3.4)	-	-	-	-	-	-	-	
	UV/VIS	In-house method* IR (KBr)		UV/VIS (Ethanol solution) UV/VIS absorption characteristics $\lambda_{\text{max}} = 271.6 \text{ nm}$ $\epsilon = 6.91 \times 10^3$ Optical purity Not applicable because pyriproxyfen is achiral Data generated were consistent with the	There is no absorption of pyriproxyfen in the visible region. Therefore, no visible spectrum was determined				

Subsection (Annex Point)	Method	Purity/ Specification	Results	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
NMR	NMR (¹ H-NMR: 270 MHz)		chemical structure Data generated were consistent with the chemical structure					
MS	MS (Electron impact ionisation)	_	Data generated were consistent with the chemical structure					
	IR (KBr), NMR (¹ H-NMR: 300 MHz proton spectrum), MS (direct infusion, positive ion electrospray)		Data generated using IR, NMR and MS were consistent with the chemical structure	-				

Section A3 Physical and Chemical Properties of Active Substance

Section	ion A5	Thysical and Chemical Troperties of Active Substance							
	Subsection (Annex Point)	Method	Purity/ Specification	Results	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
3.5	Solubility in water (IIA3.5)	EPA CG-1500*1 EEC A6, OECD 105		In the neutral range In distilled water: 0.367mg/l at 25°C In the acidic range (pH 4 to 6) and in the alkaline range (pH 8 to 10) Not required as pyriproxyfen is not capable of forming ions because of its extremely low solubility in water. Mean water solubility at 20 ± 0.5°C pH 5: 0.058 mg/L pH 7: 0.101 mg/L pH 9: 0.119 mg/L					
3.6	Dissociation constant (-)	OECD 112 (Titration method or conductometric method or spectrophotometric method)	-	Dissociation constant Dissociation constant of pyriproxyfen could not be obtained because the solubility of pyriproxyfen in water was extremely low.	-				
3.7	Solubility in organic	CIPAC MT181*		<i>n</i> -Heptane: 25 - 29 g/L at 20°C	-1				

Beet	ion A3	rnysical and Chemical Properties of Active Substance								
	Subsection (Annex Point)	Method	Purity/ Specification	Results	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only	
	solvents, including the effect of temperature on solubility (IIIA3.1)			1,2-Dichloroethane: >1000 g/L at 20°C Methanol: 25 - 29 g/L at 20°C Acetone: >1000 g/L at 20°C p-Xylene: >1000 g/L at 20°C Ethyl acetate: >1000 g/L at 20°C						
3.8	Stability in organic solvents used in b.p. and identity of relevant breakdown products (IIIA3.2)	-	-	-	Not required as the active substance as manufactured does not contain an organic solvent	-	-	-		
3.9	Partition coefficient n-octanol/water (IIA3.6)	OECD 107 equivalent to EEC A.8 – Shake flask method EEC A8, OECD 117		n-octanol/water partition co-efficient log Pow = 5.37 at 25°C (pH 5.60-5.65) Effect of pH (4 to 10) on the n-octanol/water partition co-efficient Not required as pyriproxyfen is not capable of forming ions because of its extremely low solubility in water. Mean Log P _{ow} pH 5: 4.85 pH 7: 4.86 pH 9: 4.87						

	ion A3	Thysical and Chemical Properties of Active Substance								
	Subsection (Annex Point)	Method	Purity/ Specification	Results	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only	
				Results demonstrate that n-octanol/ water partition co-efficient is not affected by pH.						
3.10	Thermal stability, identity of relevant breakdown products (IIA3.7)	-	-1	-	Refer to IIIB, 3.7 Pyriproxyfen is stable in the product when stored for 6 months at 40°C and 3 months at 60°C.	-	-	-		
3.11	Flammability, including auto-flammability and identity of combustion products (IIA3.8)	EEC A.10		Flammability: Not highly flammable	-					
		EEC A.12		Flammability (contact with water): No flammable gasses evolved	-					
		EEC A.15 (The melting point of the active substance is below 50°C)		Auto-flammability: No auto-ignition observed below 400°C, with sample volumes up to 1 mL, at an atmospheric pressure of 97.5 kPa.	-					

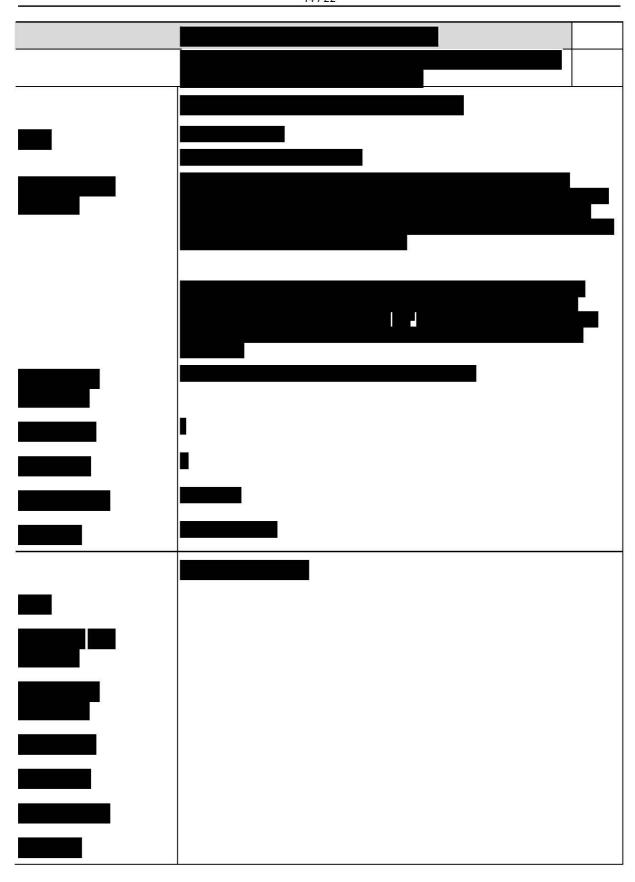
~~~	ion A5	TARES A STORES THEE. BEING THE VERLENCES A	acui i i oper iies	of Active Substance				-	
	Subsection (Annex Point)	Method	Purity/ Specification	Results	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
3.12	Flash-point (IIA3.9)	-	-	-	Not required as the melting point of pyriproxyfen is not below 40°C.	-	-	-	
3.13	Surface tension (IIA3.10)	-	-	-	Not required as the solubility of the active substance in water is below lmg/L.	-	-	-	
3.14	Viscosity (-)	-	-	-	Not required as the active substance is not a liquid	-	i.a	-	
3.15	Explosive properties (IIA3.11)	Theoretical assessment - Structural considerations and DSC testing		Pyriproxyfen is considered not to be potentially explosive based on the chemical structure and associated thermodynamic properties.	The experimental procedure described in EEC A.14 was not attempted because the assessment concluded that there was no strong evidence of explosive properties				
3.16	Oxidizing properties (IIA3.12)	Theoretical assessment - Structural considerations, oxygen balance and DSC testing		Pyriproxyfen is considered not to be potentially oxidising based on the chemical structure, oxygen balance and associated thermodynamic properties.	The experimental procedure described in EEC A.17 was not attempted because the assessment concluded that there was no strong evidence of oxidising properties				

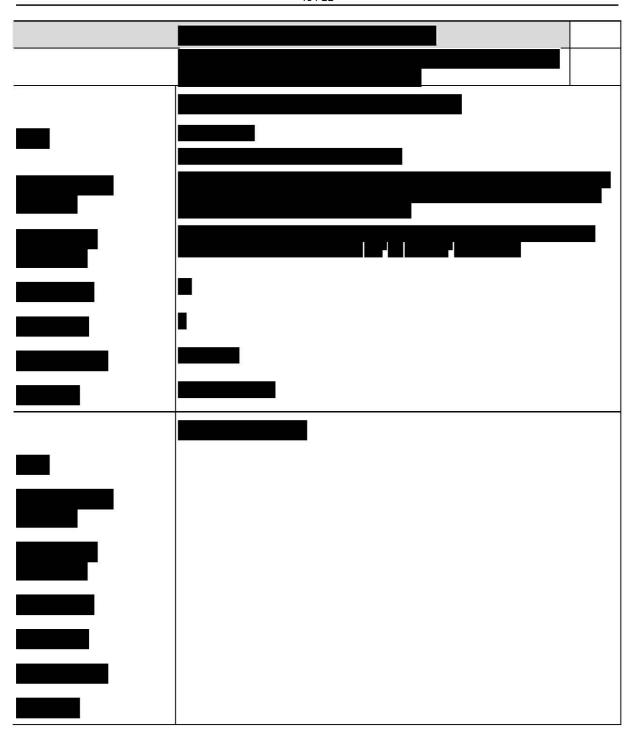
(A	Subsection Annex Point)	Method	Purity/ Specification	Results	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
coı	eactivity towards ontainer material IA3.13)	-	•	l .	No evidence of reactivity towards containers.	-	-	-	

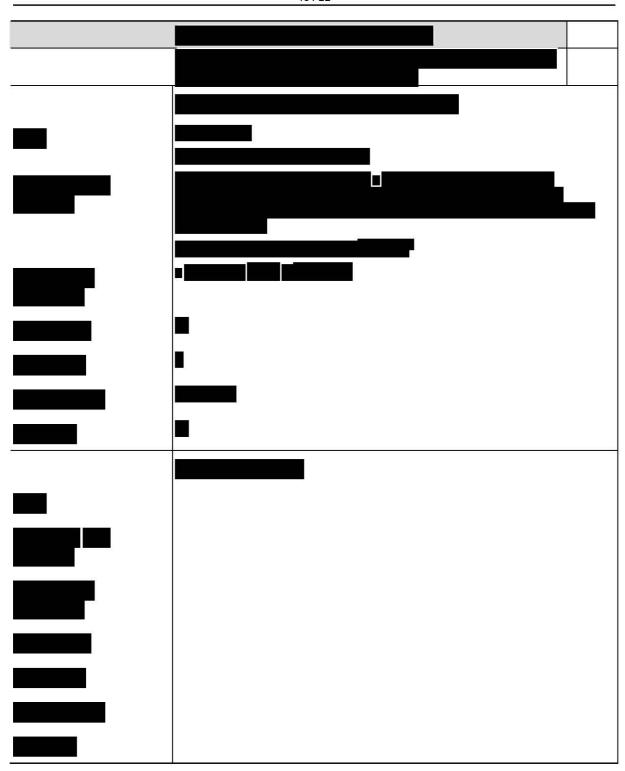
^{*} No test guideline is specified

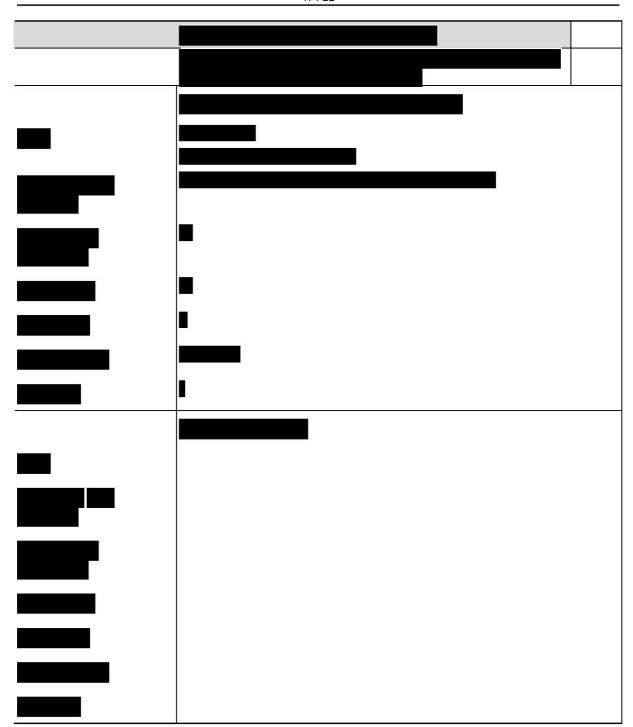


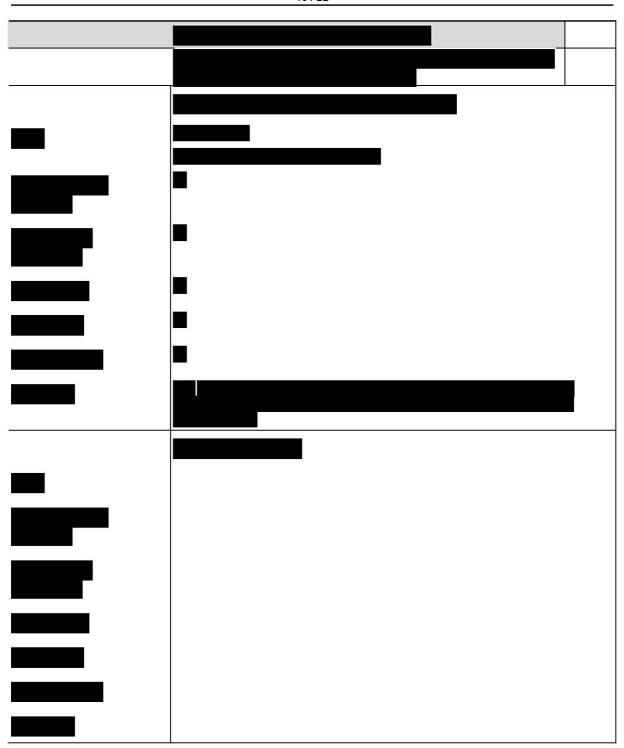


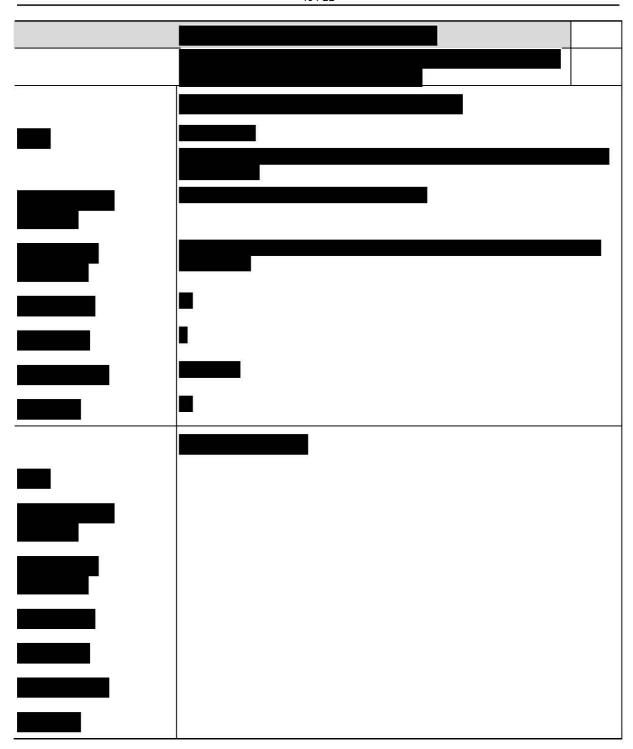


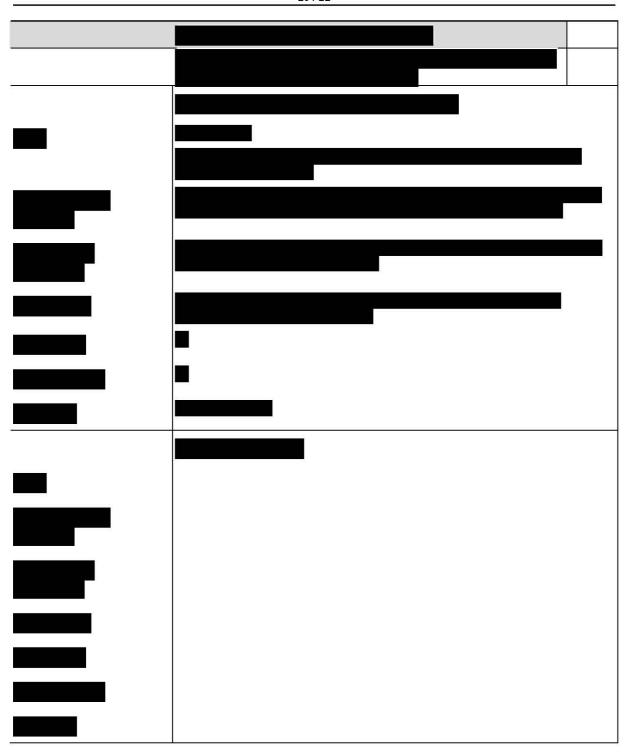


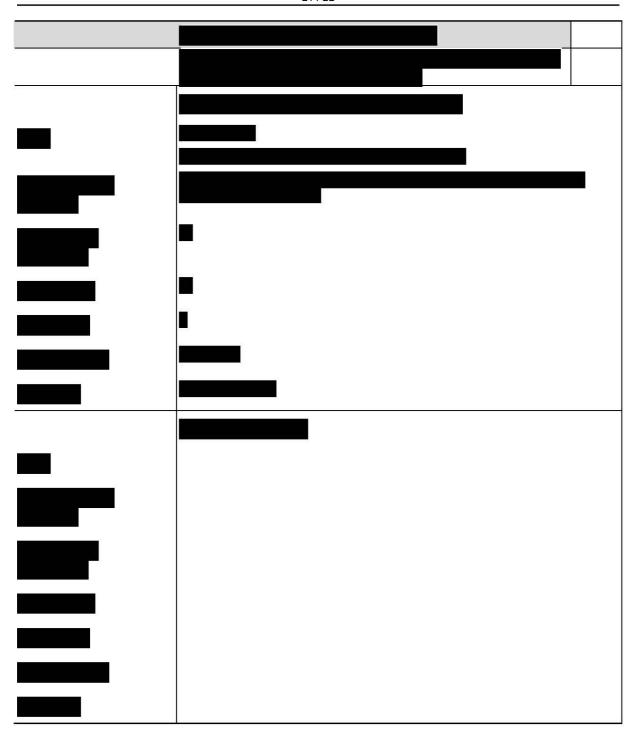


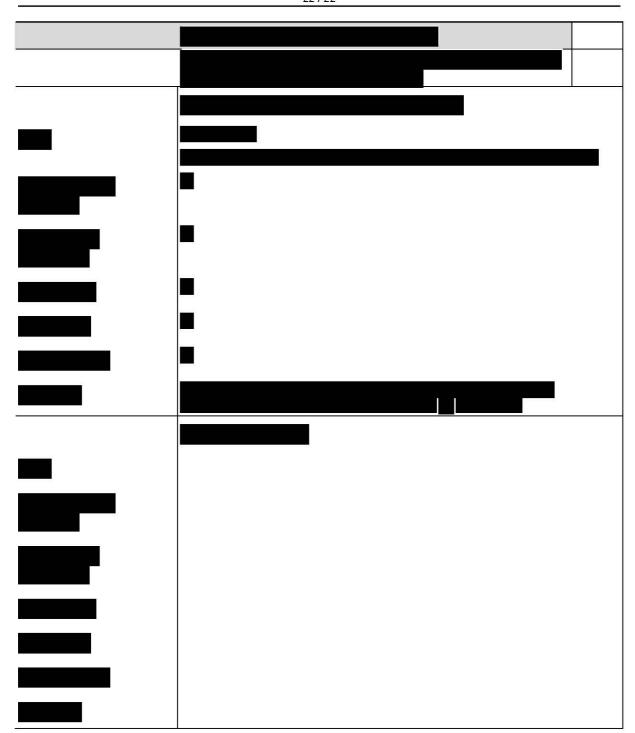












# **European Commission**



# **Pyriproxyfen**

CAS number 95737-68-1

# **Document III-A**

Section 4 Analytical Methods
Study Summaries
Active Substance

# Rapporteur Member State: The Netherlands January 2012

Draft CA-report and Proposed Decision of The Netherlands in the context of the Possible inclusion of pyriproxyfen in Annex I of Council Directive 98/8/EC

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Please refer to "Technical Notes for Guidance on Dossier Preparation including preparation and evaluation of study summaries under Directive 98/8 EC Concerning the Placing of Biocidal Products on the Market (Appendix 7.1 and 7.2)" for a list of the Standard Terms and Abbreviations used in this document.

- 4.1 Analytical methods for the determination of pure active substances and, where appropriate, for relevant degradation products, isomers and impurities of active substances and their additives (e.g. stabilisers)
- 4.1.1 Methods for the determination of pure active substance in the active substance as manufactured

Section A4.1.1/01,02 Methods for the determination of pure active substance in the active substance as manufactured

Annex Point IIA 4.1.1 Official 1 Reference use only 1.1 Reference Original validation report (2000)Addendum report (2002)1.2 Data protection Yes 1.2.1 Data owner Sumitomo Chemical Co., Ltd. 1.2.3 Criteria for data Data submitted to the MS after 13 May 2000 on existing a.s. for protection the purpose of its entry into Annex I 2 Guidance and Quality Assurance 2.1 **Guideline study** No (methods used comparable to SANCO/3030/99 rev.4) 2.2 **GLP** 2.3 Deviations 3 Materials and Methods 3.1 Preliminary treatment 3.1.1 Enrichment A known weight of the technical material is dissolved in internal standard solution (ca. 0.067 % w/v p-benzyldiphenyl methanol solution) Not relevant 3.1.2 Cleanup 3.2 Detection 3.2.1 Separation method Pyriproxyfen is separated by high performance liquid chromatography using a LiChrosorb RP-18 column at room temperature and a mobile phase of acetonitrile/water 3.2.2 Detector Pyriproxyfen is detected using an ultra violet absorption

	photometer (HPLC-UV) at a wavelength of 254 nm		
3.2.3 Standard(s)	Pyriproxyfen standards in p-benzyldiphenyl methanol (internal)		

3.2.4 Interfering substance(s) Internal standard, solvent, impurities

# 3.3 Linearity

### 3.3.1 Calibration range

Concentrations of pyriproxyfen in internal standard solution (ca. 0.067 % w/v p-benzyldiphenyl methanol solution) were prepared by accurately weighing 40, 45, 50, 55 and 60 mg pyriproxyfen and dissolving in 10 mL of internal standard solution equivalent to 4.0, 4.5, 5.0, 5.5 and 6.0 mg/mL (corresponding to 80, 90, 100, 110 and 120% of the nominal concentration in the method i.e. 5.0 mg/mL). Internal standard was added equivalent to 0.67 mg/mL

# 3.3.2 Number of measurements

5 solutions at different concentration were prepared

### 3.3.3 Linearity

Results showed the response of the HPLC-UV to be linear when the peak area ratio of pyriproxyfen/internal standard was plotted against the concentration of the pyriproxyfen standard. Using the method of least squares the correlation coefficient was calculated to be 0.99996 (slope 0.01781, Y intercept 0.01289)

The equation for the calibration line is as follows:

y = 0.01781 x + 0.01289

y: peak area ratio of pyriproxyfen to the internal standard x: concentration of pyriproxyfen (%)

# 3.4 Specifity: interfering substances

A solution containing 5.0 mg/mL pyriproxyfen standard, 0.67 mg/mL internal standard and 10  $\mu$ g/mL of each of impurities #2 and #6 (equivalent to 0.2% nominal concentration), 20  $\mu$ g/mL of impurity #3 (equivalent to 0.4% nominal concentration), 100  $\mu$ g/mL of impurity #4 (equivalent to 2.0% nominal concentration) and 25 $\mu$ g/mL of impurity #5 (equivalent to 0.5% nominal concentration) were analysed according to the analytical method. An example chromatogram presented in the report shows good specificity with no interference from solvent and complete resolution between neighbouring peaks

The identity of the peak assigned as pyriproxyfen was confirmed 2002 using HPLC-MS positive electrospray ionisation (ESI +ve) and showing that the mass spectrum of the peak identified in pyriproxyfen technical was identical to that of the pyriproxyfen standard

# 3.5 Recovery rates at different levels

Not performed. Accuracy of the method was evaluated from linearity, precision and specificity.

3.5.1 Relative standard deviation

Not performed. Accuracy of the method was evaluated from linearity, precision and specificity

# 3.6 Limit of determination

Not performed. The lowest calibration standard analysed as part of the test for linearity was 4.0 mg/mL although it is expected that the LOQ of the method could be lowered if necessary

# 3.7 Precision

### 3.7.1 Repeatability

For precision, the content of pyriproxyfen in the technical material (was a second was determined at 80%, 100% and 120% of the sample amount specified in the analytical method. Each

level was determined in triplicate and the repeatability of the method evaluated

The mean value and the standard deviation of the factors, which are quotients of the amount of pyriproxyfen related to the peak area ratios were used to calculate the RSD of 0.38%. The results are presented in Table A4.1.1-01

3.7.2 Independent laboratory validation

Not performed

# 4 Applicant's Summary and conclusion

### 4.1 Materials and methods

**Method** – Dissolving in internal standard in methanol. Quantification by HPLC-UV at 254 nm

**Specificity** – Good resolution of the active substance from the internal standard and other impurities. No interference observed. Peak identity confirmed by HPLC-MS

**Linearity** - Across the range of 80 –120% of the nominal concentration expected, the response of the HPLC-UV was found to be linear when the peak area ratio of pyriproxyfen/internal standard was plotted against the concentration of pyriproxyfen. Correlation coefficient was calculated to be 0.99996

**Accuracy** – Was based on linearity, precision and specificity of the method

Precision – 0.38% RSD, no outliers removed

### 4.2 Conclusion

The analytical methods have been validated in terms of specificity, linearity, accuracy and precision and are therefore considered suitable for the determination of pyriproxyfen in the technical material.

There are no existing CIPAC methods for pyriproxyfen.

4.2.1 Reliability

4.2.2 Deficiencies

# Evaluation by Competent Authorities Use separate "evaluation boxes" to provide transparency as to the comments and views submitted Evaluation by Rapporteur Member State Date Materials and Methods Results and discussion Conclusion

	07.25		
Reliability			
Acceptability			
Remarks			
	Comments from		
Date			
Materials and Methods			
Results and discussion			
Conclusion			
Reliability			
Acceptability			
Remarks			

	-	

# 4.1.2 Methods for the determination of significant and/or relevant impurities and additives (eg. stabilisers) in the active substance as manufactured

There are no additives in the active substance as manufactured. Please refer to the Confidential Appendix for methods for the determination of significant and/or relevant impurities in the active substance as manufactured

# 4.2 Analytical methods in all relevant environmental media including recovery rates and the limits of determination for the active substance, and for residues thereof, and where relevant in/on the following

The studies carried out for the methods for the determination of residues are summarised below:

Matrix	Method	Limit of quantification	Reference
Soil	GC-NPD	0.01 mg/kg	2002a (
Drinking water	GC-NPD	0.1 μg/L	2002
Surface water	GC-NPD	0.01 μg/L	(2001
Air	GC-NPD	0.04 μg/m³	(2001

NPD=Nitrogen phosphorus detector, MSD=Mass selective detector

Based on the validation results achieved it is considered that the methods indicated in the following text are acceptable for post registration surveillance for pyriproxyfen in accordance with EU guidance document SANCO/825/00

#### 4.2.1 Residues in soil

Section A4.2.1/01 Residues in soil

**Annex Point IIA 4.2.1** 

-				
		1 Reference	Official use only	
1.1 Re	ference	2002		
1.2 Da	ta protection	Yes		
1.2.1 Г	ata owner	Sumitomo Chemical Co., Ltd.		
1.2.3 C	riteria for data	Data submitted to the MS after 13 May 2000 on existing a.s. for		
protect	1011	the purpose of its entry into Annex I		
		2 Guidance and Quality Assurance		
2.1	Guideline study	Yes, EU Guidance document SANCO/825/00 rev.6		
2.2	GLP			
2.3	Deviations			
		3 Materials and Methods		
3.1 Pro	eliminary treatment			
ac ac		Pyriproxyfen is extracted from soil samples by shaking with acetone. Water was added beforehand in an amount that takes full account of the natural water content of the samples so that during extraction the acetone/water ratio remains constant at 2/1 (v/v)		
3.1.2 C	leanup	Following extraction, an aliquot of the filtrate is taken into separating funnel and sodium chloride and cyclohexane/ethyl		

acetate=1/1 (v/v) is added. After repeated mixing excess water was separated. The evaporation residue of the organic phase is then applied to gel chromatography on polystylene gel Bio-Beads S-X3 and eluted with a mixture of cyclohexane/ethyl acetate=1/1 (v/v). The residue purified by gel chromatography is applied to silica gel with mixtures of hexane/toluene and toluene/acetone for clean-up. The fraction containing pyriproxyfen is then concentrated prior to analysis

#### 3.2 Detection

#### 3.2.1 Separation method

Primary separation is by gas chromatography using a capillary column, DB-17 (18 m x 0.18 mm x 0.3  $\mu m$  film) in conjunction with a pre-column of deactivated fused silica (1 m x 0.25 mm)

Confirmatory separation is by gas chromatography using a capillary column, DB-17 ( $20 \text{ m} \times 0.18 \text{ mm} \times 0.3 \text{ } \mu\text{m}$  film) in conjunction with a pre-column of deactivated fused silica ( $1 - 2 \text{ m} \times 0.32 \text{ mm}$ )

#### 3.2.2 Detector

Primary detection is by gas chromatography with a nitrogen phosphorous detector (GC-NPD)

Confirmation detection is by gas chromatography with mass spectrometry (GC-MS, EI, positive) using ion m/z 226 for quantification and ions m/z 186, 136 and 96 as qualifier ions

#### 3.2.3 Standard(s)

Pyriproxyfen standards in toluene (external)

#### 3.2.4 Interfering substance(s)

Soil co-extractive material

#### 3.3 Linearity

#### 3.3.1Calibration range

Concentrations of pyriproxyfen in toluene were prepared at 0.01, 0.02, 0.05, 0.1, 0.2, 0.5 and 1.0  $\mu$ g/mL

## 3.3.2 Number of measurements

7 solutions at different concentration were prepared

#### 3.3.3 Linearity

The response of the GC-NPD and GC-MS to pyriproxyfen over the range 0.01 to  $1~\mu g/ml$  was shown to be linear although correlation was performed using a least squares fit of a non-linear function. The correlation coefficient was 0.9995 using GC-NPD and 0.9996 using GC-MS.

## 3.4 Specifity: interfering substances

Specificity was measured by the analysis of control samples and confirmation of peak identity and recovery by GC-MS

The method was shown to be specific with no interference being observed in control samples (<30% LOQ 0.01 mg/kg). The confirmatory method was adequately shown as the determination was performed using gas chromatography with a mass selective detector (MSD, quantification ion 226 and confirmation ions 96, 136 and 186)

## 3.5 Recovery rates at different levels

Validation of the method was performed by analysing control and fortified samples of soil (Speyer 2.3, Germany, sandy loam). Control soil was fortified with pyriproxyfen at 0.01and 0.10 mg/kg and then subjected to the analytical method.

The recovery values from soil at individual fortification levels

ranged from 104.1 to 107.2% (mean 105.0%, n = 5) at 0.01 mg/kg and 92.3 to 110.7% (mean 102.6%, n = 5) at 0.1 mg/kg. The overall mean recovery was 103.8% (n = 10)

Recovery confirmation using GC-MS ranged from 105.9 to 109.5% (mean 107.3%, n=3) at 0.01 mg/kg and 102.4% (n=1) at 0.1 mg/kg

The maximum time between starting the extraction and measurement was 9 days (16 days for GC-MS). Extracts were stored in a refrigerator before GC measurement. Results indicate there is no impact due to this storage period

3.5.1 Relative standard deviation

The relative standard deviation at individual fortification levels was 1.2% (n = 5) at 0.01 mg/kg and 7.5% (n = 5) at 0.1 mg/kg. The overall relative standard deviation was 5.2% (n = 10)

RSD confirmation using GC-MS was 1.8% (n = 3) at 0.01 mg/kg

3.6 Limit of determination

The LOQ was determined successfully at 0.01 mg/kg. An LOD was approximately 0.004 mg/kg estimated from the lowest calibration standard  $(0.01 \, \mu g/mL)$ 

3.7 Precision

3.7.1 Repeatability

Refer to 3.5.1

3.7.2 Independent laboratory validation

#### 4 Applicant's Summary and conclusion

#### 4.1 Materials and methods

**Method** – Extraction with acetone, clean up by liquid-liquid partition, GPC, followed by silica gel purification. Quantification by GC-NPD. Confirmation by GC-MS

**Specificity** - No interference being observed in control samples (<30% LOQ 0.01 mg/kg). Peak identity and recovery confirmed by GC-MS

**Linearity** - The response of the GC-NPD and GC-MS to pyriproxyfen over the range 0.01 to 1.0  $\mu$ g/mL was linear although correlation was performed using a least squares fit of a non-linear function. The correlation coefficient ( $r^2$ ) was typically >0.9995

**Accuracy** - Overall mean recovery was 104%, n = 10

**Precision** - Overall RSD was 5.2%, n = 10

Limit of quantification - LOQ 0.01 mg/kg

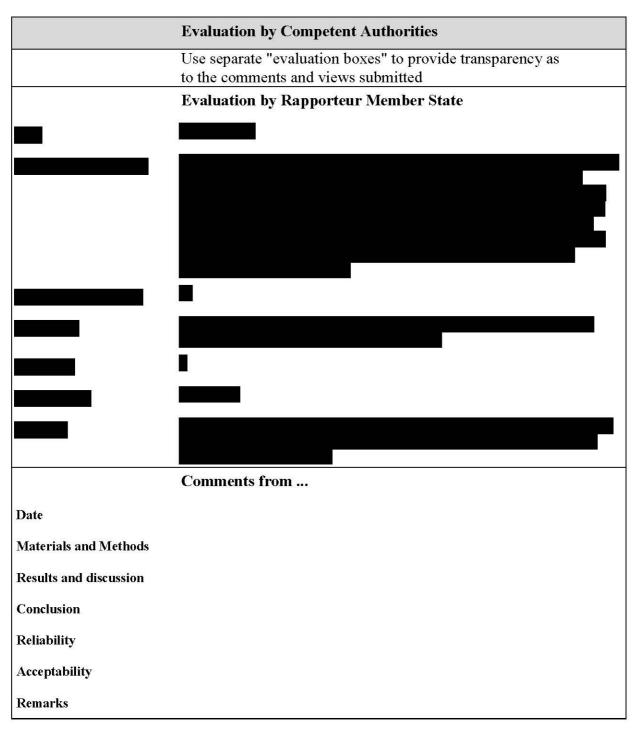
#### 4.2 Conclusion

The analytical method has been validated in terms of specificity, linearity, accuracy and precision and are therefore considered suitable for the determination of pyriproxyfen in soil

The limit of quantification for the analytical method for residues of pyriproxyfen in soil is lower than the concentration of concern in exposure to non-target soil organisms. The results of the ecotoxicological studies of non-target soil organisms are as follows:

Earthworm: LC50 > 1000 mg/kg soil, NOEC 1000 mg/kg soil Soil non-target micro-organisms: NOEC 14 mg/kg soil Therefore, the method is suitable for the determination of

4.2.1 Reliability		
4.2.2 Deficiencies		





#### Residues in water (including drinking water and surface water) 4.2.2

Residues in water (including drinking water and **Section A4.2.2/01** 

Annex Point IIA 4.2.2	curfo co water)		
	1 Reference	Official use only	
1.1 Reference	2002		
1.2 Data protection	Yes		
1.2.1 Data owner	Sumitomo Chemical Co., Ltd.		
1.2.3 Criteria for data protection	Data submitted to the MS after 13 May 2000 on existing a.s. for the purpose of its entry into Annex I		
	2 Guidance and Quality Assurance		
2.1 Guideline study	Yes, EU Guidance document SANCO/825/00 rev.6		
2.2 GLP			
2.3 Deviations			
	3 Materials and Methods		
3.1 Preliminary treatment			
3.1.1 Enrichment	The water sample is passed through a solid phase extraction (SPE) C18 column under negative pressure. The SPE C18 column is then washed with 10 ml distilled water and dried by suction of air through the column for approximately 10 minutes. Pyriproxyfen is then eluted with 20 ml acetone. The eluate is evaporated to dryness and dissolved in toluene to an appropriate volume		
3.1.2 Cleanup	Not relevant		
3.2 Detection			
3.2.1 Separation method	Primary separation is by gas chromatography using a capillary column, DB-17 (18 m x 0.18 mm x 0.3 $\mu$ m film) in conjunction with a pre-column of deactivated fused silica (1 m x 0.25 mm) Confirmatory separation is by gas chromatography using a capillary column, DB-17 (20 m x 0.18 mm x 0.3 $\mu$ m film) in conjunction with a pre-column of deactivated fused silica (1 - 2 m x 0.32 mm)		

Pyriproxyfen: CAS number 95737-68-1 Doc.IIIA – Section 4 Analytical methods – Active substance January 2012 RMS: NL

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#### 3.2.2 Detector

Primary detection is by gas chromatography with a nitrogen

phosphorous detector (GC-NPD)

Confirmation detection is by gas chromatography with mass spectrometry (GC-MS, EI, positive) using ion m/z 226 for quantification and ions m/z 186, 136 and 96 as qualifier ions

3.2.3 Standard(s)

Pyriproxyfen standards in toluene (external)

3.2.4 Interfering substance(s)

Drinking water co-extractive material

#### 3.3 Linearity

3.3.1 Calibration range

Concentrations of pyriproxyfen in toluene were prepared at 0.01, 0.02, 0.05, 0.1, 0.2, 0.5 and 1.0  $\mu$ g/mL

3.3.2 Number of measurements

7 solutions at different concentration were prepared

3.3.3 Linearity

The response of the GC-NPD and GC-MS to pyriproxyfen over the range 0.01 to 1  $\mu$ g/ml was shown to be linear although correlation was performed using a least squares fit of a non-linear function. The correlation coefficient was 0.9983 using GC-NPD and 0.9994 using GC-MS

## 3.4 Specifity: interfering substances

Specificity was measured by the analysis of control samples and confirmation of peak identity and recovery by GC-MS

The method was shown to be specific with no interference being observed in control samples (<30% LOQ  $0.1~\mu g/L$ ). The confirmatory method was adequately shown as the determination was performed using gas chromatography with a mass selective detector (MSD, quantification ion 226 and confirmation ions 96, 136 and 186)

### 3.5 Recovery rates at different levels

Validation of the method was performed by analysing control and fortified samples of drinking water (regular laboratory tap water). Control water was fortified with pyriproxyfen at 0.1and 1.0 µg/L and then subjected to the analytical method.

The recovery values from water at individual fortification levels ranged from 97.4 to 111.0% (mean 104.2%, n = 5) at 0.1  $\mu$ g/L and 74.8 to 83.5% (mean 78.9%, n = 5) at 1.0  $\mu$ g/L. The overall mean recovery was 91.6% (n = 10)

Recovery confirmation using GC-MS ranged from 91.8 to 94.2% (mean 92.6%, n=3) at 0.1  $\mu$ g/L and 83.7% (n=1) at 1.0  $\mu$ g/L

The maximum time between starting the extraction and measurement was 7 days (15 days for GC-MS). Extracts were stored in a refrigerator before GC measurement. Results indicate there is no impact due to this storage period

3.5.1 Relative standard deviation

The relative standard deviation at individual fortification levels was 5.4% (n = 5) at 0.1  $\mu$ g/L and 4.3% (n = 5) at 1.0  $\mu$ g/L. The overall relative standard deviation was 15.3% (n = 10).

RSD confirmation using GC-MS was 1.5% (n = 3) at 0.1  $\mu$ g/L

#### 3.6 Limit of determination

The LOQ was determined successfully at 0.1  $\mu$ g/L. An LOD was approximately 0.02  $\mu$ g/L estimated from the lowest calibration

standard (0.01 µg/mL)

#### 3.7 Precision

3.7.1 Repeatability Refer to 3.5.1
3.7.2 Independent laboratory Not performed validation

#### 4 Applicant's Summary and conclusion

#### 4.1 Materials and methods

**Method** – Extraction with SPE C18 column. Quantification by GC-NPD. Confirmation by GC-MS.

Specificity - No interference being observed in control samples (<30% LOQ 0.1 µg/L). Peak identity and recovery confirmed by GC-MS

**Linearity** - The response of the GC-NPD and GC-MS to pyriproxyfen over the range 0.01 to 1.0  $\mu$ g/mL was linear although correlation was performed using a least squares fit of a non-linear function. The correlation coefficient (r²) was typically >0.998

**Accuracy** - Overall mean recovery was 91.6%, n = 10

**Precision** - Overall RSD was 15.3%, n = 10

Limit of quantification -  $LOQ~0.1~\mu g/L$ 

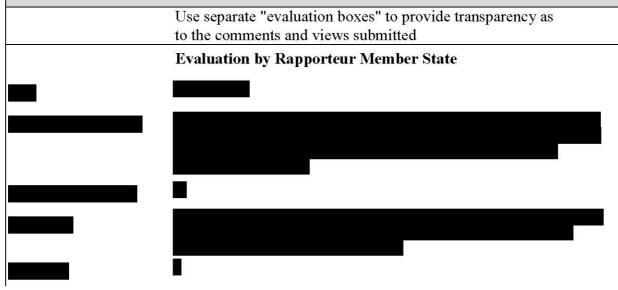
#### 4.2 Conclusion

The analytical method has been validated in terms of specificity, linearity, accuracy and precision and has been successfully validated down to the EU limit of 0.1  $\mu\text{g/L}$ . Therefore, the method is considered appropriate for post registration surveillance for pyriproxyfen in drinking water

4.2.1 Reliability

4.2.2 Deficiencies

#### **Evaluation by Competent Authorities**



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	Comments from
Date	
Materials and Methods	
Results and discussion	
Conclusion	
Reliability	
Acceptability	
Remarks	

4			

#### **Section A4.2.2/02**

## Residues in water (including drinking water and surface water)

#### **Annex Point IIA 4.2.2** Official 1 Reference use only 1.1 Reference 2001 1.2 Data protection Yes 1.2.1 Data owner Sumitomo Chemical Co., Ltd. 1.2.3 Criteria for data Data submitted to the MS after 13 May 2000 on existing a.s. for protection the purpose of its entry into Annex I 2 Guidance and Quality Assurance 2.1 Yes, EU Guidance document SANCO/825/00 rev.6 Guideline study 2.2 GLP 2.3 **Deviations** 3 Materials and Methods 3.1 Preliminary treatment 3.1.1 Enrichment The water sample is passed through a solid phase extraction (SPE) C18 column under negative pressure. The SPE C18 column is then washed with 10 ml distilled water and dried by suction of air through the column for approximately 10 minutes. Pyriproxyfen is then eluted with 10 ml hexane. The eluate is evaporated to dryness and dissolved in toluene to an appropriate volume Not relevant 3.1.2 Cleanup 3.2 Detection 3.2.1 Separation method Primary separation is by gas chromatography using a capillary column, DB-17 (20 m x 0.18 mm x 0.3 µm film) in conjunction with a pre-column of deactivated fused silica (1 m x 0.25 mm) Confirmatory separation is by gas chromatography using a capillary column, DB-17 (20 m x 0.18 mm x 0.3 µm film) in conjunction with a pre-column of deactivated fused silica (1 - 2 m x 0.32 mm3.2.2 Detector Primary detection is by gas chromatography with a nitrogen phosphorous detector (GC-NPD) Confirmation detection is by gas chromatography with mass spectrometry (GC-MS, EI, positive) using ion m/z 226 for quantification and ions m/z 186, 136 and 96 as qualifier ion 3.2.3 Standard(s) Pyriproxyfen standards in toluene (external)

3.2.4 Interfering substance(s)	Surface water co-extractive material
--------------------------------	--------------------------------------

#### 3.3 Linearity

3.3.1 Calibration range Concentrations of pyriproxyfen in toluene were prepared at 0.01,

0.02, 0.05, 0.1, 0.2 and 0.5 μg/mL

3.3.2 Number of measurements

6 solutions at different concentration were prepared

3.3.3 Linearity

The response of the GC-NPD and GC-MS to pyriproxyfen over the range 0.01 to  $0.5~\mu g/ml$  was shown to be linear although correlation was performed using a least squares fit of a non-linear function for GC-MS. The correlation coefficient was 0.9999 using GC-NPD and 0.991 using GC-MS

## 3.4 Specifity: interfering substances

Specificity was measured by the analysis of control samples and confirmation of peak identity and recovery by GC-MS

The method was shown to be specific with no interference being observed in control samples (<30% LOQ 0.01 µg/L). The confirmatory method was adequately shown as the determination was performed using gas chromatography with a mass selective detector (MSD, quantification ion 226 and confirmation ions 96, 136 and 186)

## 3.5 Recovery rates at different levels

Validation of the method was performed by analysing control and fortified samples of surface water (water from River Kleine Wiese, Germany). Control water was fortified with pyriproxyfen at 0.01 and 0.10  $\mu$ g/L and then subjected to the analytical method. The results are presented in Table A4.2.2-02

The recovery values from water at individual fortification levels ranged from 106.8 to 108.5% (mean 107.7%, n = 5) at 0.01 µg/L and 89.0 to 106.0% (mean 97.5%, n = 5) at 0.10 µg/L. The overall mean recovery was 102.6% (n = 10).

Recovery confirmation using GC-MS was 100.7% (n = 1) at 0.10 µg/L.

The maximum time between starting the extraction and measurement was 1 day (11 days for GC-MS). Extracts were stored in a freezer before GC measurement. Results indicate there is no impact due to this storage period

3.5.1 Relative standard deviation

The relative standard deviation at individual fortification levels was 0.6% (n = 5) at  $0.01~\mu g/L$  and 7.1% (n = 5) at  $0.10~\mu g/L$ . The overall relative standard deviation was 7.0% (n = 10)

#### 3.6 Limit of determination

The LOQ was determined successfully at 0.1  $\mu$ g/L. An LOD was approximately 0.005  $\mu$ g/L estimated from the lowest calibration standard (0.01  $\mu$ g/mL)

#### 3.7 Precision

3.7.1 Repeatability

Refer to 3.5.1

3.7.2 Independent laboratory validation

#### 4 Applicant's Summary and conclusion

#### 4.1 Materials and methods

**Method** – Extraction with SPE C18 column. Quantification by

GC-NPD. Confirmation by GC-MS

Specificity - No interference being observed in control samples (<30% LOQ 0.01  $\mu$ g/L). Peak identity and recovery confirmed by GC-MS

**Linearity** - The response of the GC-NPD and GC-MS to pyriproxyfen over the range 0.01 to 1.0  $\mu$ g/mL was linear although correlation was performed using a least squares fit of a non-linear function for GC-MS. The correlation coefficient ( $r^2$ ) was typically 0.9999 for GC-NPD and 0.991 for GC-MS.

**Accuracy** - Overall mean recovery was 102.6%, n = 10

**Precision** - Overall RSD was 7.0%, n = 10

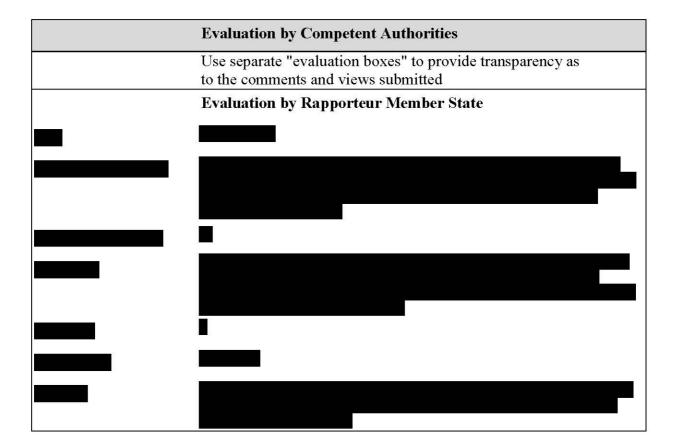
Limit of quantification -  $LOQ~0.01~\mu\text{g/L}$ 

4.2 Conclusion

The analytical method has been validated in terms of specificity, linearity, accuracy and precision and the limit of quantification for the method for the residues of pyriproxyfen in surface water is lower than the concentration of concern for exposure to the most sensitive species in aquatic non-target organisms (Daphnia magna NOEC (21 days) 0.015  $\mu g/L$ ). Therefore, the method is suitable for the determination of pyriproxyfen in surface water for monitoring purposes

4.2.1 Reliability

4.2.2 Deficiencies



## Comments from ... Date **Materials and Methods** Results and discussion Conclusion Reliability Acceptability Remarks

-			

#### 4.2.3 Residues in air

#### **Section A4.2.3/01**

#### Residues in air

#### **Annex Point IIA 4.2.3**

#### Official 1 Reference use only 1.1 Reference 2001 1.2 Data protection Yes 1.2.1 Data owner Sumitomo Chemical Co., Ltd. 1.2.3 Criteria for data Data submitted to the MS after 13 May 2000 on existing a.s. for protection the purpose of its entry into Annex I 2 Guidance and Quality Assurance Yes, EU Guidance document SANCO/825/00 rev.6 and BBA 2.1 Guideline study 'Richtlinien fur die Prufung von Pflanzenschutzmitteln im Zulassungsverfahren Teil I (1-2) Seite B/8, 2.5, 2. Auflage, November 1990 auf erstmalige/erneute Zulassung eines Pflanzenschutzmittels 2.2 GLP 2.3 **Deviations** 3 Materials and Methods 3.1 Preliminary treatment 3.1.1 Enrichment Tenax adsorption tubes were placed in the air suction apparatus. The flow rate was adjusted to 1 L/min for each sample resulting in a sampling volume of about 360 L after a sampling period of 6 h. Pyriproxyfen was extracted from tenax adsorption tubes with 10 ml of toluene. The extract was filtrated through 0.45 μm filter prior to analysis Not relevant 3.1.2 Cleanup 3.2 Detection 3.2.1 Separation method Primary separation is by gas chromatography using a capillary column, DB-17 (20 m x 0.18 mm x 0.3 µm film) in conjunction with a pre-column of deactivated fused silica (1 m x 0.25 mm) Confirmatory separation is by gas chromatography using a capillary column, DB-17 (20 m x 0.18 mm x 0.3 µm film) 3.2.2 Detector Primary detection is by gas chromatography with a nitrogen phosphorous detector (GC-NPD)

Confirmation detection is by gas chromatography with mass spectrometry (GC-MS, EI, positive) using ion m/z 226 for quantification and ions m/z 186, 136 and 96 as qualifier ions.

RMS: NL

3.2.3 Standard(s) Pyriproxyfen standards in toluene (external)

3.2.4 Interfering substance(s) Absorbent material and co-extractives from air

3.3 Linearity

3.3.1 Calibration range Concentrations of pyriproxyfen in toluene were prepared at 0.01, 0.02, 0.05, 0.1, 0.2, 0.5 and 1.0 μg/mL

3.3.2 Number of 7 solutions at different concentration were prepared measurements

3.3.3 Linearity The response of the GC-NPD and GC-MS to pyriproxyfen over

the range 0.01 to 1.0 µg/ml was shown to be linear although correlation was performed using a least squares fit of a non-linear function for GC-MS. The correlation coefficient typically in the range 0.995 to 0.999 using GC-NPD and 0.997 using GC-MS

3.4 Specifity: interfering substances

Specificity was measured by the analysis of control samples and confirmation of peak identity and recovery by GC-MS.

The method was shown to be specific with no interference being observed in control samples ( $<30 \% LOQ 1 \mu g/m^3$ ). The confirmatory method was adequately shown as the determination was performed using gas chromatography with a mass selective detector (MSD, quantification ion 226 and confirmation ions 96, 136 and 186)

#### 3.5 Recovery rates at different levels

Validation of the method was performed by fortifying traps containing Tenax absorbent and drawing air through the traps at 1 L/min for 6 hours (total 360 L) to simulate sampling conditions. Two different air sampling conditions were used. One set of traps sampled air at 20°C and relative humidity 30% whilst a second set of traps sampled air at 35°C and relative humidity >80%. For each set of sampling conditions, control traps were fortified in quintuplet with pyriproxyfen to simulate air concentrations of 1 and 10 µg/m³. Secondary traps were attached to the main traps fortified at the higher concentration and were analysed to check for breakthrough of pyriproxyfen from the main trap

Results showed good accuracy for both sets of samples. No breakthrough was observed from the main trap.

The recovery values from Tenax adsorption tubes at individual fortification levels ranged from:

88.8 to 94.4% (mean 91.3%, n = 5) at 1.0  $\mu$ g/m³ and 95.4 to 107.0% (mean 101.1%, n = 5) at  $10 \,\mu\text{g/m}^3$  for air at  $20^{\circ}\text{C}$ , 30%RH (overall mean 96.2%, n = 10)

 $107.2 \text{ to } 109.0\% \text{ (mean } 108\%, n = 5) \text{ at } 1.0 \text{ }\mu\text{g/m}^3 \text{ and } 107.1 \text{ to }$ 108.7% (mean 108.0%, n = 5) at 10  $\mu$ g/m³ for air at 35°C, 80% RH (overall mean 108.0%, n = 10).

Recovery confirmation using GC-MS was 104.1% (n = 1) at 1.0 $\mu g/m^3$ 

The maximum time between starting the extraction and measurement was 7 days (14 days for GC-MS). Extracts were stored in a freezer before GC measurement. Results indicate there is no impact due to this storage period

## 3.5.1 Relative standard deviation

Results showed good repeatability for both sets of samples. The RSD values for Tenax adsorption tubes at individual fortification levels ranged from:

2.8 to 4.9% for air at 20°C, 30% RH (overall RSD 6.6%, n = 10) and 0.6 to 0.7% for air at 35°C, 80% RH (overall RSD 0.6%, n = 10)

#### 3.6 Limit of determination

The LOQ was determined successfully at 1  $\mu$ g/m³. An LOD was approximately 0.3  $\mu$ g/m³ estimated from the lowest calibration standard (0.01  $\mu$ g/mL)

#### 3.7 Precision

3.7.1 Repeatability

Refer to 3.5.1

3.7.2 Independent laboratory validation

#### 4 Applicant's Summary and conclusion

#### 4.1 Materials and methods

**Method** – Extraction of Tenax absorbent with toluene. Quantification by GC-NPD. Confirmation by GC-MS

**Specificity** – No interference observed in control samples (<30% LOQ 1  $\mu g/m^3$ ). Confirmation of peak identity and recovery by GC-MS

**Linearity** - The response of the GC-NPD and GC-MS to pyriproxyfen over the range 0.01 to 1.0  $\mu$ g/ml was shown to be linear although correlation was performed using a least squares fit of a non-linear function. The correlation coefficient ( $r^2$ ) was typically > 0.995

**Accuracy** - Overall mean recovery was 96.2% for air sampled at  $20^{\circ}$ C, RH 30% and 108% for air sampled at  $35^{\circ}$ C, RH >80%

**Precision** – Overall RSD was 6.6% for air sampled at 20°C, RH 30% and 0.6% for air sampled at 35°C, RH >80%

Limit of quantification - LOQ 1 µg/m³

**Breakthrough** - No breakthrough of pyriproxyfen from the main traps

#### 4.2 Conclusion

The analytical method has been validated in terms of specificity, linearity, accuracy and precision and the limit of quantification for the analytical method for residues of pyriproxyfen in air is 1  $\mu g/m^3$  which is sufficiently below the concentration C defined in the SANCO/825/00 rev.6, taking into account relevant health based limit values and relevant exposure levels. Therefore, the method is suitable for the determination of pyriproxyfen in air for monitoring purposes

#### 4.2.1 Reliability

4.2.2 Deficiencies

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	Evaluation by Competent Authorities
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted
	<b>Evaluation by Rapporteur Member State</b>
	Comments from
Date	
Materials and Methods	
Results and discussion	
Conclusion	
Reliability	
Acceptability	
Remarks	

#### 4.2.4 Residues in body fluids and tissues

No methods are required since pyriproxyfen is not classified as toxic or highly toxic

#### 4.3 Residues in/on food or feedstuffs and other products where relevant

No methods are required since the product is not intended for use on food or feedstuffs

## **European Commission**



**Pyriproxyfen** 

**CAS number 95737-68-1** 

Document III-A Study Summaries Active Substance Efficacy

# Rapporteur Member State: The Netherlands January 2012

Draft CA-report and Proposed Decision of The Netherlands in the context of the Possible inclusion of Pyriproxyfen in Annex I of Council Directive 98/8/EC