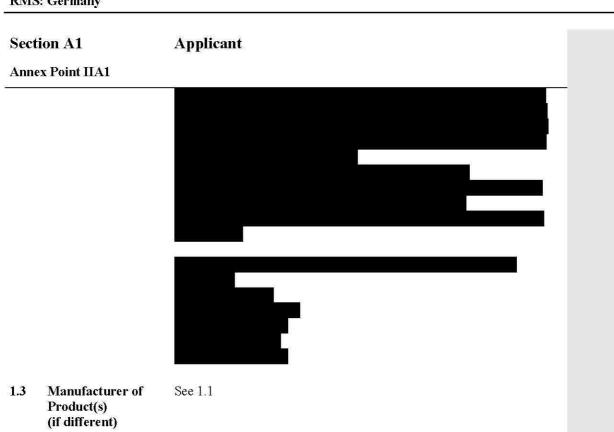
Applicant Section A1 Annex Point IIA1 1.1 **Applicant** 1.2 Manufacturer of **Active Substance** (if different)



	Evaluation by Competent Authorities
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted
	EVALUATION BY RAPPORTEUR MEMBER STATE
Date	2014/03/04
Materials and methods	
Conclusion	
Reliability	I
Acceptability	
Remarks	-
	COMMENTS FROM
Date	Give date of comments submitted
Results and discussion	Discuss additional relevant discrepancies referring to the (sub)heading numbers and to applicant's summary and conclusion. Discuss if deviating from view of rapporteur member state
Conclusion	Discuss if deviating from view of rapporteur member state
Reliability	Discuss if deviating from view of rapporteur member state

July 2007

Section A2

Identity of Active Substance

	section ex Point)					Official use only	
2.1	Common name (IIA2.1)	2-Propanol	2-Propanol				
2.2	Chemical name (IIA2.2)	Propan-2-ol					
2.3	Manufacturer's development code number(s) (IIA2.3)	not applicable					
2.4	CAS No and EC numbers (IIA2.4)						
2.4.1	CAS-No	67-63-0					
2.4.2	EC-No	200-661-7 (EINE	CS)				
2.4.3	Other	EC No: 603-117-0 RTECS No: NT80				X	
2.5	Molecular and structural formula, molecular mass (IIA2.5)						
2.5.1	Molecular formula	C_3H_8O					
2.5.2	Structural formula	HO CH 3 CH 3 CC(OH)C (= SMI	LES code)				
2.5.3	Molecular mass	60.09 g/mol					
2.6	Method of manufacture of the active substance (IIA2.1)	process. In this pro 60% sulfuric acid step process. Anothydration of prope	ocess propene and the result ther important ene with water atalyst, mixed	gas is absorbed in ing sulfates hydrol manufacturing pro r. Hydration can be phase with a catio	yzed in a single ocess is catalytic gas-phase with a		
2.7	Specification of the purity of the active substance, as appropriate (IIA2.7)	g/kg	g/1	% w/w 99.9	% v/v Min. 99.0 99.997	X	

Section A2

Identity of Active Substance

2.8 Identity of impurities and additives, as appropriate (IIA2.8)

Impurities of marketed propan-2-ol may vary depending on the purity grade (p.a. or technical product), on manufacturers handling and on single production specifications represented by a lot number. Impurities other than water are volatile components as well as acid contents in variable compositions.

H ₂ O	Max. 5000 ppm	
Organic acids	Max. 30 ppm	j
Other organic impurities	Max. 3000 ppm	
Non volatile matter	Max. 50 ppm	

2.8.1 Isomeric composition

not applicable

2.9 The origin of the natural active substance or the precursor(s) of the active substance (IIA2.9)

not applicable

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Section A2 Identity of Active Substance

	Evaluation by Competent Authorities
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted
	EVALUATION BY RAPPORTEUR MEMBER STATE
Date	2014/03/04
Materials and methods	
Condenie	
Conclusion	
Reliability	
Acceptability	
Remarks	e:
	COMMENTS FROM
Date	Give date of comments submitted
Results and discussion	Discuss additional relevant discrepancies referring to the (sub)heading numbers and to applicant's summary and conclusion. Discuss if deviating from view of rapporteur member state
Conclusion	Discuss if deviating from view of rapporteur member state
Reliability	Discuss if deviating from view of rapporteur member state
Acceptability	Discuss if deviating from view of rapporteur member state
Remarks	

Task Fo RMS: G	rce "2-Propanol" ermany	Propan-2-ol				July 2007
	n A2.8 Point IIA2.8	Identity of impuri fill in one form for each		12704	ve substance)	
Subse	ction					Official use only
2.8.1.1	Common name	Water				
2.8.1.2	Function	None, impurity				X
2.8.2	IUPAC name	Hydrogen dioxide				x
2.8.3	CAS-No	7732-18-5				
2.8.4 2.8.5	EC-No Other CIPAC	231-791-2 (EINECS)				
2.8.6	Molecular formula	H_2O				
2.8.7	Structural formula	Н				
2.8.8	Molecular mass	18.0153 g/mol				
2.8.9	Concentration of the impurity or additive	g/kg	g/l	% w/w <0.01	% v/v Max. 5000 ppn 0.014	X

typical and range of concentrations

	Evaluation by Competent Authorities
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted
	EVALUATION BY RAPPORTEUR MEMBER STATE
Date	2008/06/10
Materials and methods	
Conclusion	
Reliability	
Acceptability	
Remarks	-
	COMMENTS FROM
Date	Give date of comments submitted
Results and discussion	Discuss additional relevant discrepancies referring to the (sub)heading numbers and to applicant's summary and conclusion. Discuss if deviating from view of rapporteur member state
Conclusion	Discuss if deviating from view of rapporteur member state
Reliability	Discuss if deviating from view of rapporteur member state
Acceptability	Discuss if deviating from view of rapporteur member state
Remarks	

Task Force "2-Propanol"	Propan-2-ol	September 2008
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Section A3	Physical and Chem	ical Properties	of Active Substance

	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
3.1	Melting point, boiling point, relative density (IIA3.1)								
3.1.1	Melting point								
	Melting pt. 1	Handbook data	No data	result: -89.5 °C pressure: -	Peer-reviewed handbook data			Merck (1996)	X
	Melting pt. 2	Handbook data	No data	result: -89.5°C pressure: -	Peer-reviewed handbook data			CRC (2001)	X
3.1.2	Boiling point								
	Boiling pt. 1	Handbook data	No data	result: 82.5 °C pressure: 1013 hPa	Peer-reviewed handbook data		I	Merck (1996)	X
	Boiling pt. 2	Handbook data	No data	result: 82.3 °C pressure: 1013 hPa	Peer-reviewed handbook data			CRC (2001)	X
3.1.3	Bulk density/ relative density								
	Rel. density 1	Handbook data	No data	0.785 at 20°C	Peer-reviewed handbook data			Merck (1996)	X
	Rel. density 2	Handbook data	No data	0.785 at 20°C	Peer-reviewed handbook data			Sax (1984)	X
	Rel. density 3	Handbook data	No data	0.7809 at 25°C	Peer-reviewed handbook data			CRC (2001)	X

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The first control of the first	Section A3	Physical and Chemical Properties of Active Substance
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Dece	on AJ	Thysical and Chen	nem i roperne.	of Active Substance					
	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
3.2	Vapour pressure (IIA3.2)								
	Vapour pressure 1	Handbook data	No data	result: 57.8 hPa (5.775 Pa) temperature: 25 °C	Acceptable handbook data			Riddick et al (1986)	X
	Vapour pressure 2	Calculated from Riedel's regression equation	No data	result: 60.5 hPa temperature: 25 °C	Handbook data)	Daubert & Danner (1985)	X
	Vapour pressure 3	Calculated from Antoine-type equation	No data	result: 60.73 hPa temperature: 25 °C	Handbook data			Yaws (1997)	X
3.2.1	Henry's Law Constant (Pt. I-A3.2)								
	Henry's Law Const. 1	GC	No data	Measured (at 25 °C): result: 0.80 Pa · m³/mol	Study well documented, meets generally accepted scientific principles			Snider & Dawson (1985)	X
	Henry's Law Const. 2	Calculation using HENRYWIN v3.10	12	Calculated (at 25 °C): result: 0.76 Pa · m³/mol (bond method) 1.16 Pa · m³/mol (group method)	Generally accepted calculation method				X
	Henry's Law Const. 3	Handbook data	No data	result: 0.82 Pa · m³/mol	Reliable handbook data			Howard (1990)	X

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Section A3	Physical and Chemic	cal Properties of Active Substance
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	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
	Henry's Law Const. 4	No data	No data	result: 0.82 Pa · m³/mol	Only result presented			Taft et al. (1985),	X
3.3	Appearance (IIA3.3)								146
3.3.1	Physical state	Visual inspection	0 =	liquid	Peer-reviewed handbook data			Sax (1984)	X
		ASTM D4176	99.9% m/m	Liquid; clear & free from suspended matter	Certificate of analysis		, ,		X
3.3.2	Colour	Visual inspection	tu:	colorless	Peer-reviewed handbook data	I			X
		ASTM D1209	99.9% m/m	Pt-Co 0	Certificate of analysis, (Platinum- Cobalt scale 0: means that substance is colourless)				X
3.3.3	Odour	Olfactory assessment	NEO .	slight	Peer-reviewed handbook data	I			X
		Shell Method Series (SMS) 547	99.9% m/m	Initial diluted odor: 4 (= same as reference solution odour)	Certificate of analysis, (reference substance: Propan-2- ol (Isopropyl Alcohol)				X

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Sect	ion A3	Physical and Chem	nical Properties	of Active Substance					
	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
3.4	Absorption spectra (IIA3.4)								
		Handbook data	No data	For propan-2-ol as a solvent in UV/VIS spectrophotometry a cut-off point of 210 nm is given. Therefore, no absorption between 290 nm and 750 nm is expected.	Peer-reviewed handbook data	3		CRC (2001)	X
		According Eur. Pharm.	99.997%	Abs 230 nm: 0.09 Abs 250 nm: 0.01 Abs 270 nm: -0.01 Abs 290 nm: -0.01 Abs 310 nm: -0.01 Spectrum is consistent with the proposed structure. No absorption maximum >210 nm	Internal report of manufacturer to applicant				X
		UV-Vis	99.997%	Spectrum is consistent with the proposed structure. No absorption maximum >210 nm	Internal report of applicant. Brenntag sales 2-propanol of SASOL				X ²
		Lambda 35 (Perkin- Elmer), Serie-Nr. 101N2012803 incl. Software UVWinlab	99.997% 99.9% m/m	Spectrum is consistent with the proposed structure for quality from both sources	Internal report of applicant				X

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Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Officia use on
	V2.85.04		•	1	1		•	. 12
IR	liquid film	No data	3346, 3334, 2972, 2933, 2907, 2884, 2722, 2659, 2521, 2408, 2387, 2198, 1903, 1766, 1467, 1409, 1379, 1368, 1341, 1309, 1162, 1130, 1110, 954, 818, 660, 654, 490, 436 cm ⁻¹	Compilation of data			SDBS (2007)	X
	Europ.Pharm. /USP	99.997%	Spectrum is consistent with the proposed structure compared with reference spectrum	Internal report of manufacturer to applicant				Х
	FT-IR Photometer Paragon 1000 (Perkin-	99.997%	Spectrum is consistent with the proposed structure for quality from both sources	Internal report of applicant				X
	Elmer), Serie-Nr. 39543 incl. Software Spectrum BX 5.0.1	99.997%						
NMR	¹ H-NMR at 300 MHz ¹³ C-NMR at 75.5 MHz	No data	ppm in CDCl ₃ CH3: 1.22 (d) CH: 4.04 (sep) ppm in CDCl ₃ CH3: 25.14 CH: 64.50	Compilation of data			Gottlieb et al. (1997)	X
	¹ H-NMR at 399 MHz	99.997%	Spectrum is consistent with the proposed structure	Internal report of applicant	.			X

Task Force "2-Propanol"	Propan-2-ol	September 2008
RMS: Germany	and viking to be	5.P.1.m.511 = 3.55

Section A3	Physical and	Chemical Properties	of Active Substance

	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
		¹ H-NMR at 399 MHz ¹³ C-NMR at 100 MHz	99.9 %m/m	Spectrum is consistent with the proposed structure	Internal report of manufacturer to applicant				X
	MS	Handbook data	No data	45(100), 43(19), 27(17), 29(12), 41(7), 31(6), 19(6), 42(5), 44(4), 59(3)	Peer-reviewed handbook data			CRC (2001)	X
		Europ.Pharm. /USP	99.997%	Spectrum is consistent with the proposed structure	Internal report of manufacturer to applicant		1		X
		Not specified	99.997	Spectrum is consistent with the proposed structure compared to reference spectrum	Internal report of applicant. Brenntag sales 2-propanol of SASOL				X
		Not specified	99.9 %m/m	Spectrum is consistent with the proposed structure compared to reference spectrum	Internal report of manufacturer to applicant				X
3.5	Solubility in water (IIA3.5)	Handbook data	No data	result: infinitely in aqua temperature: - pH: -	Acceptable handbook data; Note: Episuite database cite this source with 1000 g/L at 25°C			Riddick et al. (1986)	X
		Handbook data	No data	Miscible	Peer-reviewed			CRC (2001)	X

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-2		565			No.		44	ng1	
	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
		Handbook data	No data	Miscible with water	handbook data Peer-reviewed handbook data			Merck (1996)	X
3.6	Dissociation constant (-)	Handbook data	No data	-3.2	Acceptable handbook data			Riddick et al. (1986)	X
3.7	Solubility in organic solvents, including the effect of temperature on solubility (IIIA3.1)	Handbook data	No data	result: miscible with acetone, alcohol and ether temperature: -	Peer-reviewed handbook data			CRC (2001)	X
3.8	Stability in organic solvents used in b.p. and identity of relevant breakdown products (IIIA3.2)			Scientifically unjustified, see Justification for non-submission.	Expert judgement	I			X

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Necet.	IOII AJ	r nysicar and enen	mem rroperties	of Active Substance					
	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
3.9	Partition coefficient n-octanol/water (IIA3.6)								
	log P _{ow} 1	GC	No data	result: 0.05 temperature: - pH: -	Generally accepted determination method without detailed documentation		•	Dillingham et al. (1973)	X
	log Pow 2	Measured, no further information	No data	result: 0.05 temperature: - pH: -	Original literature not available	2.		Hansch et al. (1995)	X
3.10	Thermal stability, identity of relevant breakdown products (IIA3.7)	-	-	Scientifically unjustified, see Justification for non-submission.	Expert judgement	I	I		
3.11	Flammability, including auto-flammability and identity of combustion products	Handbook data Handbook data	No data	result: 455°C (autoignition) result: 399°C (ignition temperature)	Peer-reviewed handbook data Peer-reviewed handbook data		ı	Sax (1984) CRC (2001)	
3.12	Flash-point (IIA3.9)					35.			
	Flash-point 1	Handbook data	No data	12°C (closed cup)	Acceptable handbook data			Riddick et al. (1986)	

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	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
	Flash point 2	Handbook data	No data	12°C	Peer-reviewed handbook data			Sax (1984)	
3.13	Surface tension (IIA3.10)								X
	Surface tension 1	Handbook data	No data	result: 20.93 mN/m temperature: 25°C	Peer-reviewed handbook data			CRC (2001)	
	Surface tension 2	Prolabo tensiometer, which employs the Wilhelmy plate principle	>99.7%	result: 21.22 mN/m temperature: 25°C result: 21.74 mN/m temperature: 20°C	Study well documented, meets generally accepted scientific principles	data		Vazquez et al. (1995)	
		Calculation of surface tension according to Vasquez et al. (1995)	1g/L	result: 71.55 mN/m temperature: 20°C	Coefficients for calculation have been read from two figures which may lead to some uncertainties				c.
	Surface tension 3	Langmuir trough and Wilhemly plate	16.7 g/L aqueous solution Spectrophoto- metric grade 99+%	result: 62 mN/m temperature: Room temperature	Study well documented, meets generally accepted scientific principles. Value given as log mole fraction: -2.3			Kataoka & Cremer (2006)	

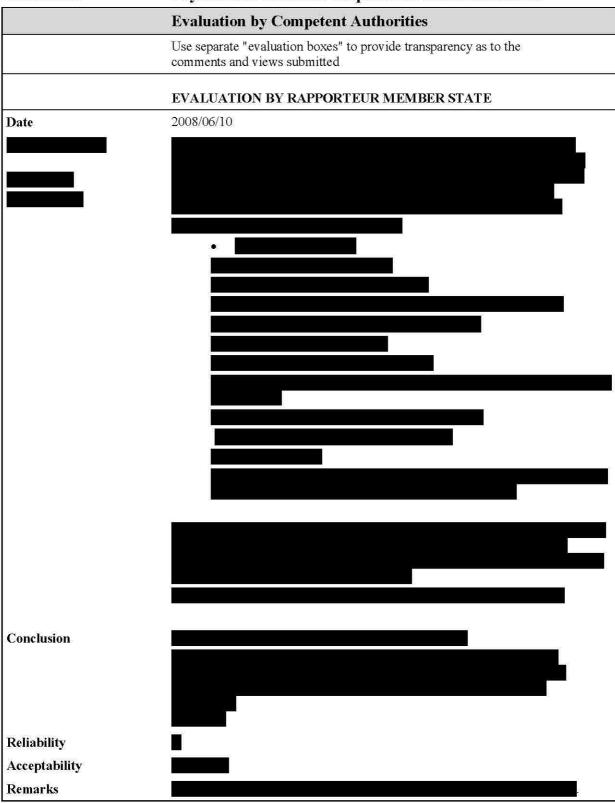
Task Force "2-Propanol"	Propan-2-ol	September 2008
RMS: Germany		2000 - 100 -

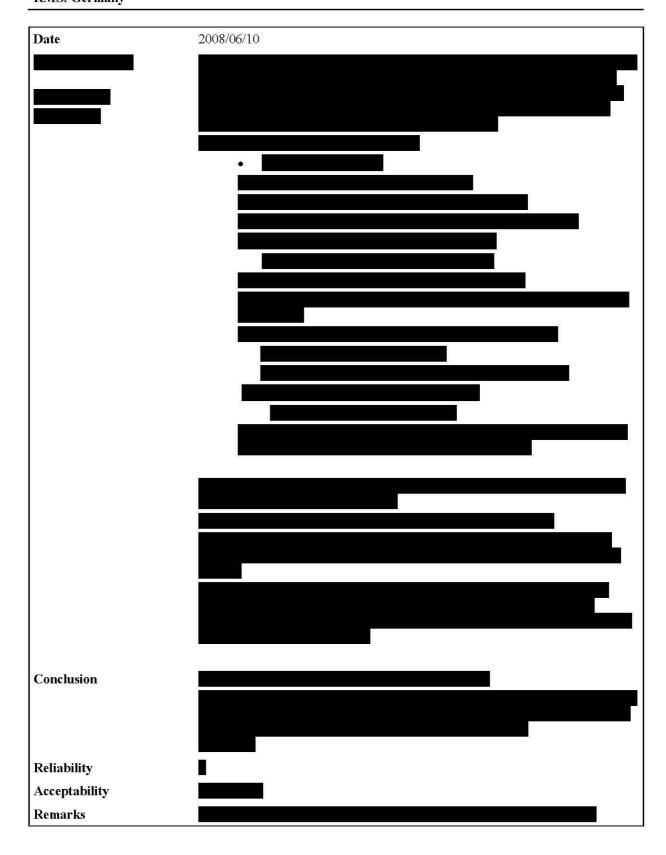
	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
	Surface tension 4	Interpolation Calculation of surface	1 g/L 1 g/L	result: 72.11 mN/m result: 72.73 mN/m	Value derived by linear interpolation of surface tension of a 16.7 g/L solution and pure water (72.75 mN/m) Values for pure			Tahery &	
		tension according to Tahery & Modarress (2005)		(ideal case) temperature: 20°C result: 72.72 mN/m (incl. deviation form ideal case) temperature: 20°C	substance and water taken from Vasquez et al. 1995			Modarress (2005),	
	Surface tension 5	OECD 115; EU A.5	1g/L, 99.7% (m/m)	result: 70.7 mN/m temperature: 22 °C					X
3.14	Viscosity (-)	Handbook data	No data	result:2.038 mPA · s temperature: 25°C	Peer-reviewed handbook data			CRC (2001)	X
3.15	Explosive properties (IIA3.11)	Handbook data	No data	Lower explosive limit in air: 2.0% Upper explosive limit in air: 12.7%	Peer-reviewed handbook data			CRC (2001)	
3.16	Oxidizing properties (IIA3.12)			Scientifically unjustified, see Justification for non-submission.	Expert judgement				

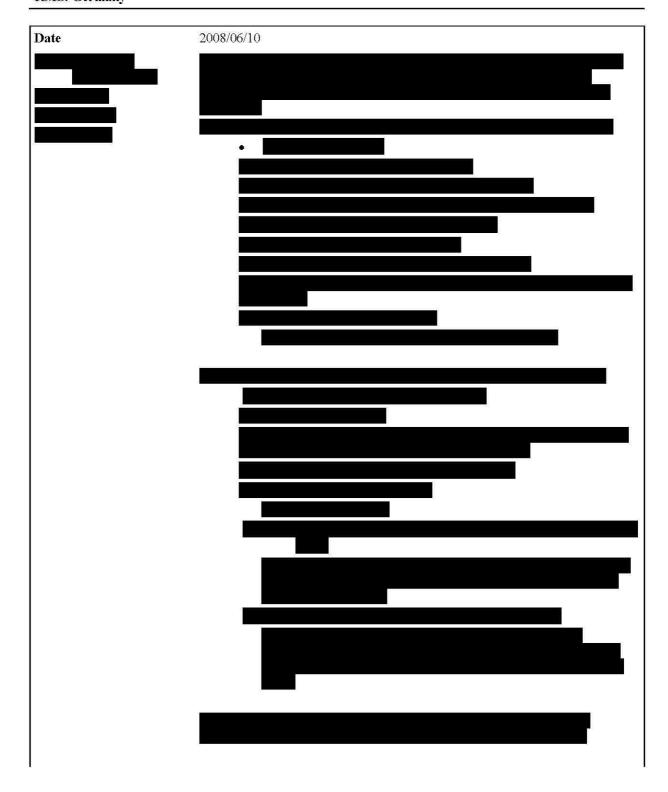
Task Force "2-Propanol"	Propan-2-ol	September 2008
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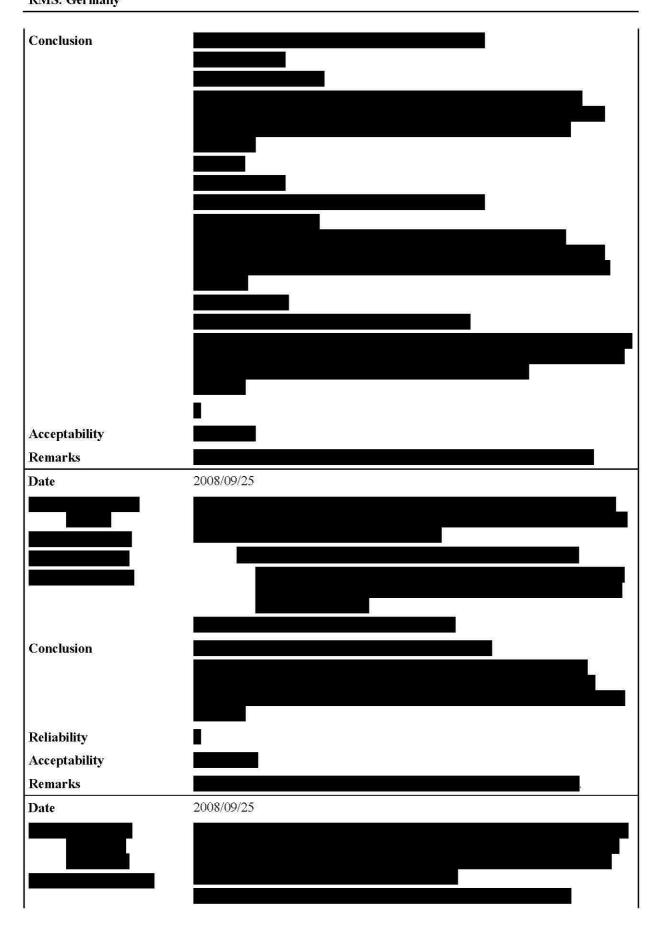
	Subsection (Annex Point)	Method	Purity/ Specification	Results Give also data on test pressure, temperature, pH and concentration range if necessary	Remarks/ Justification	GLP (Y/N)	Reliability	Reference	Official use only
3.17	Reactivity towards container material (IIA3.13)			Scientifically unjustified, see Justification for non-submission.	Expert judgement	I			

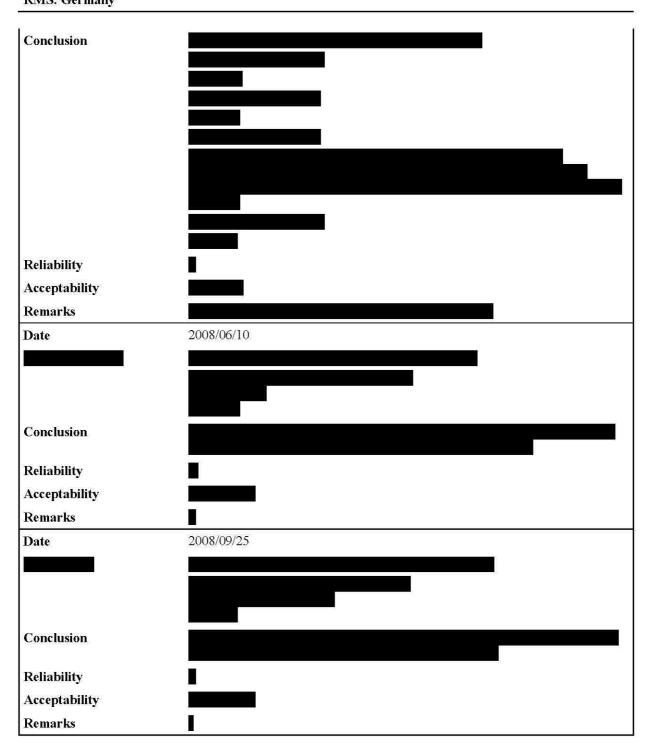
Section A3 Physical and Chemical Properties of Active Substance

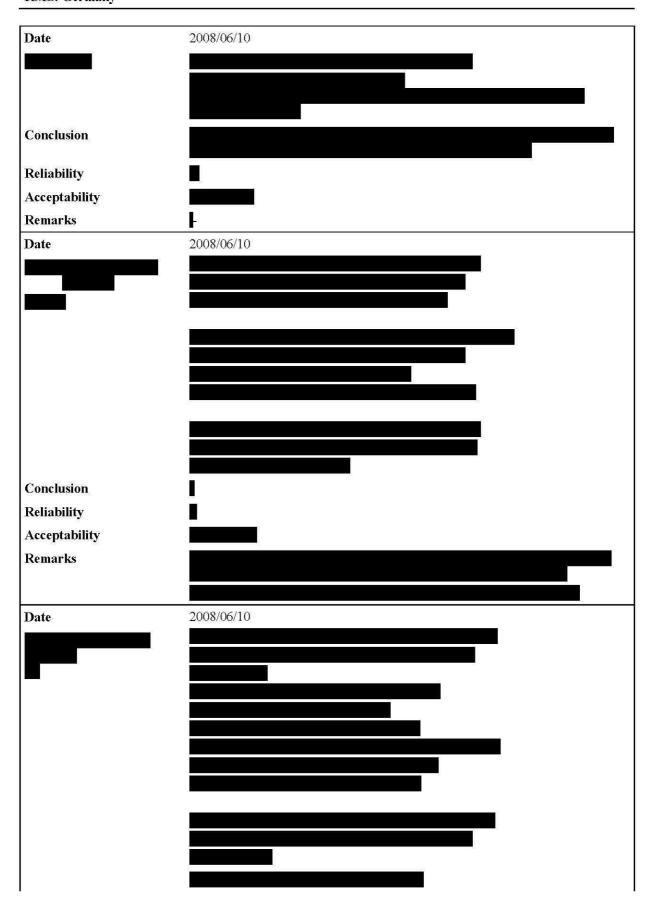


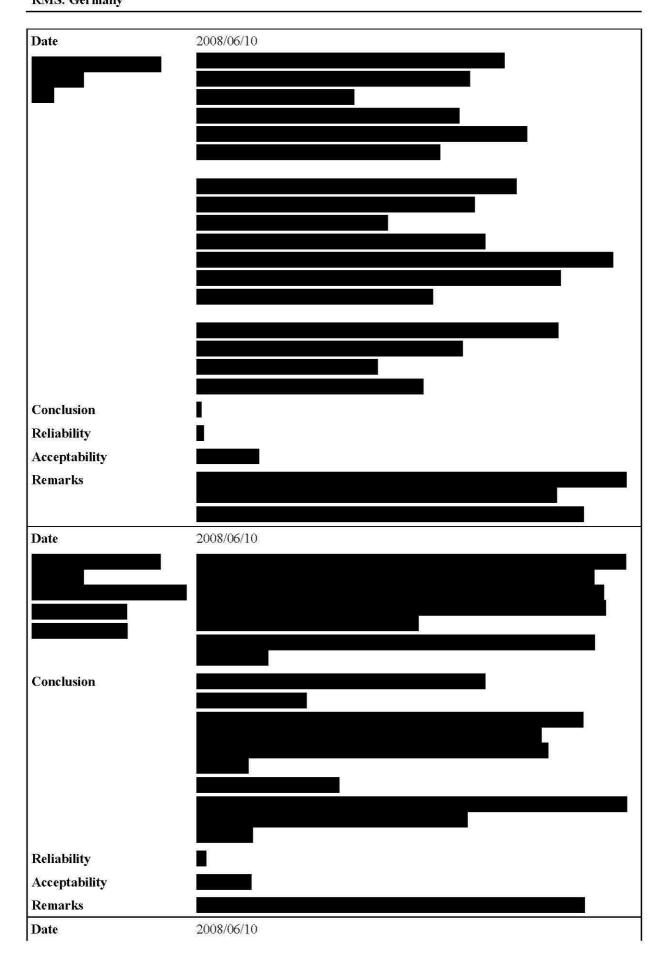


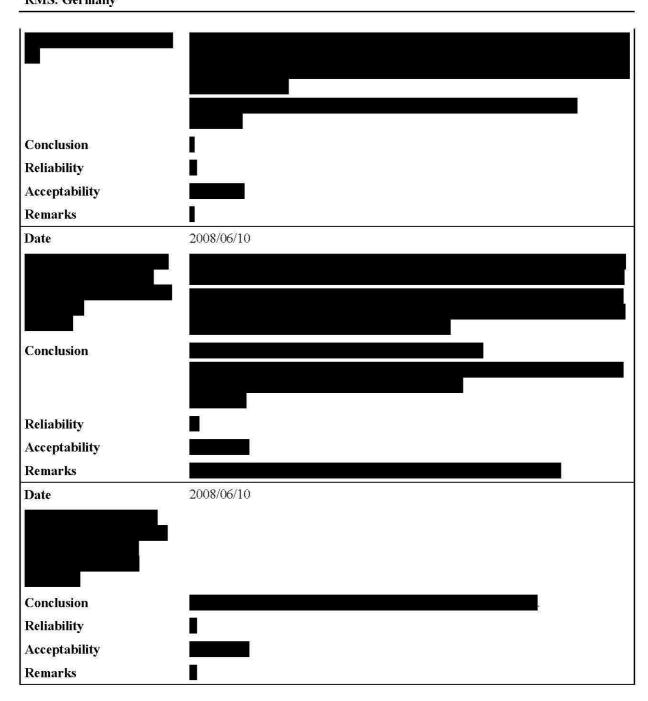


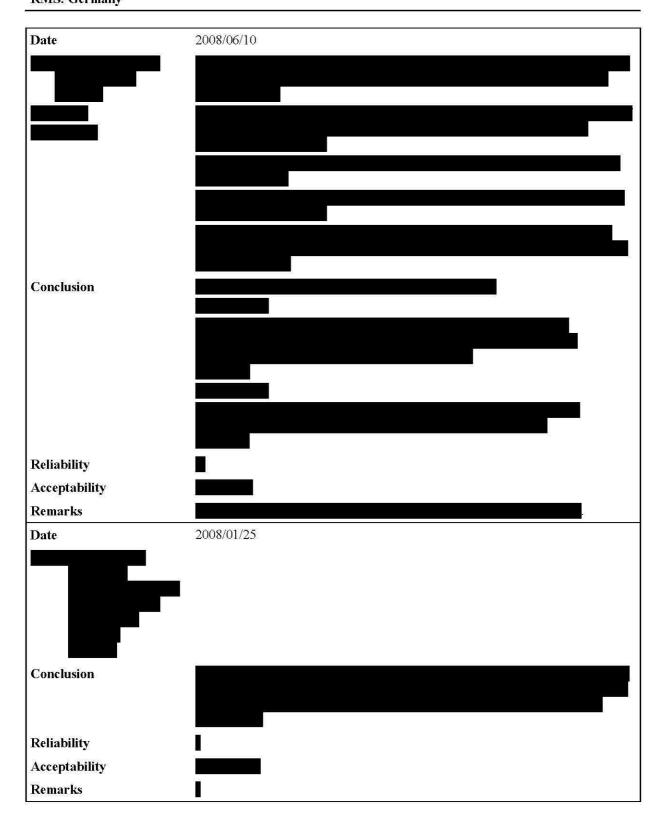


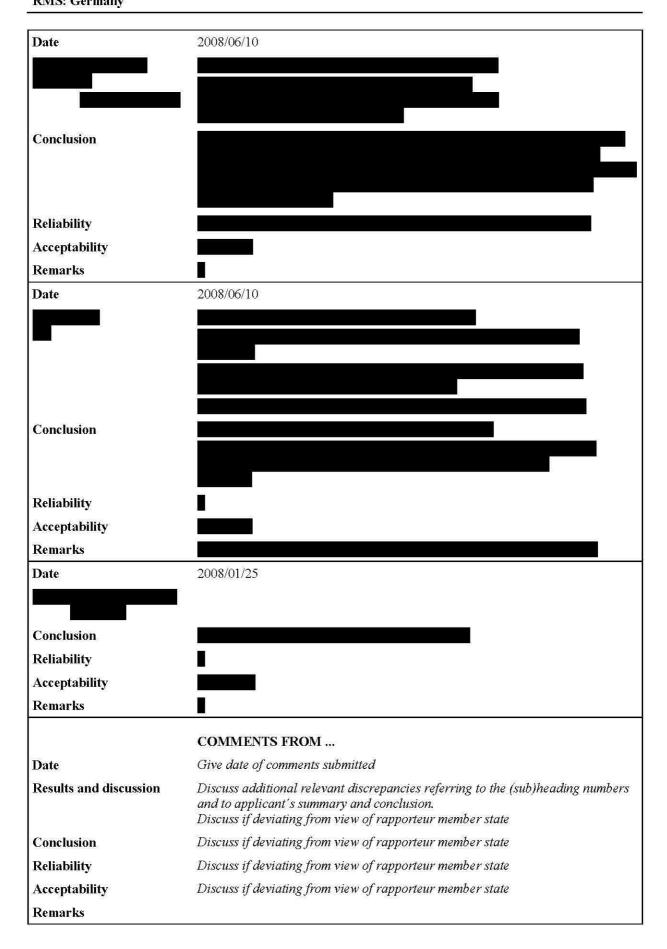












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Section A3.6 Dissociation constant Annex Point IIA3

	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Official use only
Other existing data []	Technically not feasible [] Scientifically unjustified [X]	
Limited exposure []	Other justification [X]	
Detailed justification:		
		f
		-2
References:		
	Evaluation by Competent Authorities	
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted	
	EVALUATION BY RAPPORTEUR MEMBER STATE	
Date	2008/09/25	
Evaluation of applicant's justification		
Conclusion		
Remarks	I	
	COMMENTS FROM OTHER MEMBER STATE (specify)	
Date	Give date of comments submitted	
Evaluation of applicant's justification	Discuss if deviating from view of rapporteur member state	
Conclusion	Discuss if deviating from view of rapporteur member state	
Remarks		

Remarks

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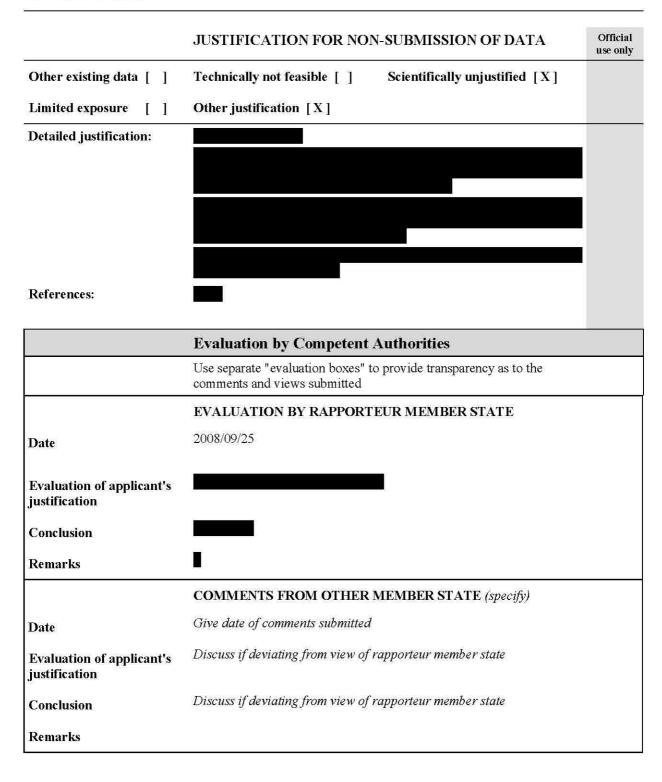
Section A3.16 Oxidizing properties

Annex Point IIA3.12

	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Official use only
Other existing data []	Technically not feasible [] Scientifically unjustified [X]	
Limited exposure []	Other justification [X]	
Detailed justification:		
		1
		_
References:		
	Evaluation by Competent Authorities	
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted	
	EVALUATION BY RAPPORTEUR MEMBER STATE	
Date	2008/09/25	
Evaluation of applicant's justification		
Conclusion		
Remarks		
	COMMENTS FROM OTHER MEMBER STATE (specify)	
Date	Give date of comments submitted	
Evaluation of applicant's justification	Discuss if deviating from view of rapporteur member state	
Conclusion	Discuss if deviating from view of rapporteur member state	
Remarks		

Section A3.17 Reactivity towards container material

Annex Point IIA3.13



Task Force "Propan-2-ol" **July 2007** Propan-2-ol RMS: Germany

Section A4.1

Analytical Methods for Detection and Identification

Annex Point IIA4.1		Determination of components in propan-2-ol			
7				Official	
		1	REFERENCE	use only	
1.1	Reference	Council of Europe (2005) EUROPEAN PHARMACOPOEIA 5.0 Monograph Isopropyl alcohol. 01/2005: 0970; p 1841-1842 (published)			
1.2	Data protection	No			
1.2.1	Data owner	(<u>C</u>)			
1.2.2	Criteria for data protection	12:			
		2	GUIDELINES AND QUALITY ASSURANCE		
2.1	Guideline study	No			
2.2	GLP				
2.3	Deviations	X 			
		3	MATERIALS AND METHORS		
3.1	Preliminary	3	MATERIALS AND METHODS		
3.1	treatment				
3.1.1	Enrichment	Not ap	Not applicable		
3.1.2	Cleanup	Not applicable			
3.2	Detection	GC/FI	D		
3.2.1	Separation method	a.	Test solution (a): substance to be determined (propan-2-ol)		
		b.	Test solution (b): 1.0 mL of 2-butanol were diluted to 50.0 mL with the test substance. 5.0 mL of this solution was further diluted to 100.0 mL with test solution (a)		
		C.	Reference solution (a): 2-butanol (0.5 mL) and 0.5 mL of propanol were diluted to 50.0 mL with the test substance. 5.0 mL of this solution were further diluted to 50 mL with test substance (a)		
		d.	Reference solution (b): $100~\mu L$ of benzene were diluted to $100.0~mL$ with the test substance. $0.20~\mu l$ of this solution were further diluted to $100.0~mL$ with the test substance (a)		
		GC conditions			
		- Material: fused silica			
		- Size	30 m in length, 0.32 mm in diameter		
		- Stationary phase: poly[(cyanopropyl)(phenyl)][dimethyl]siloxane (film thickness: 1.8 μm)			
		Carrier gas: helium for chromatography Auxiliary gas: nitrogen for chromatography or helium for Chromatography: linear velocity: 35 cm/s; split ratio: 1:5; temperature column: 40°C (0-12 min), 40-240°C (12-32 min), 240°C (32-42 min); injection port: 280°C; detector: 280°C; injection volume: 1 µL			
3.2.2	Detector	FID			
3.2.3	Standard(s)	2-butanol and benzene			
3.2.4	Interfering	No data			

Task Force "Propan-2-ol"	Propan-2-ol	July 2007
RMS: Germany		

Section A4.1 Analytical Methods for Detection and Identification

Annex Point IIA4.1

Determination of components in propan-2-ol

	Point 11A4.1	30 900 50
	substance(s)	
3.3	Linearity	
3.3.1	Calibration range	No data
3.3.2	Number of measurements	No data
3.3.3	Linearity	No data
3.4	Specifity: interfering substances	No data
3.5	Recovery rates at different levels	No data
3.5.1	Relative standard deviation	No data
3.6	Limit of	System suitability (reference solution b)
	determination	Resolution: minimum 10 between the peak (propanol) and the second peak (2-butanol) Limits
		-benzene (test solution (a)): not more than half the area of the corresponding peak in the chromatogram obtained with reference solution (b) (2 ppm), after the sensitivity has been adjusted so that the height of the peak due to benzene in the chromatogram obtained with reference solution (b) represents at least 10% of the full scale of the recorder
		-total of other impurities apart from 2-butanol (test solution (b)): not more than 3 times the area of the peak due to 2-butanol in the chromatogram obtained with test solution (b) (0.3%), after the sensitivity has been adjusted so that the height of the 2 peaks following the principle peak in the chromatogram obtained with reference solution (a) represents at least 50% of the full scale of the recorder
3.7	Precision	No data
3.7.1	Repeatability	No data
3.7.2	Independent laboratory validation	No data
		4 APPLICANT'S SUMMARY AND CONCLUSION
4.1	Materials and methods	A method for the determination of possible impurities occurring in propan-2-ol is described. For the detection a glass capillary gas chromatography method coupled with FID was developed.
4.2	Conclusion	
4.2.1	Reliability	
4.2.2	Deficiencies	

Task Force "Propan-2-ol"	Propan-2-ol	July 2007
RMS: Germany		

Section A4.1 Analytical Methods for Detection and Identification Annex Point IIA4.1 Determination of components in propan-2-ol

	Evaluation by Competent Authorities
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted
	EVALUATION BY RAPPORTEUR MEMBER STATE
Date	2008/06/10
Materials and methods	
Conclusion	ì
Reliability	
Acceptability	
Remarks	<u> </u>
	COMMENTS FROM
Date	Give date of comments submitted
Results and discussion	Discuss additional relevant discrepancies referring to the (sub)heading numbers and to applicant's summary and conclusion. Discuss if deviating from view of rapporteur member state
Conclusion	Discuss if deviating from view of rapporteur member state
Reliability	Discuss if deviating from view of rapporteur member state
Acceptability	Discuss if deviating from view of rapporteur member state
Remarks	

Task Force "2-Propanol RMS: Germany	Propan-2-ol	July 2007
Section A4.1 Annex Point IIA4.1	Analytical Methods for Detection and Identification Determination of components in propan-2-ol	
	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Official use only
Other existing data [X]	Technically not feasible [] Scientifically unjustified []	
Limited exposure []	Other justification [X]	

References:

	Evaluation by Competent Authorities
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted
	EVALUATION BY RAPPORTEUR MEMBER STATE
Date	2013/09/04
Evaluation of applicant's justification	
Conclusion	
Remarks	
	Comments from other Member State (specify)
Date	Give date of comments submitted
Evaluation of applicant's justification	Discuss if deviating from view of rapporteur member state
Conclusion	Discuss if deviating from view of rapporteur member state
Remarks	

Propan-2-ol

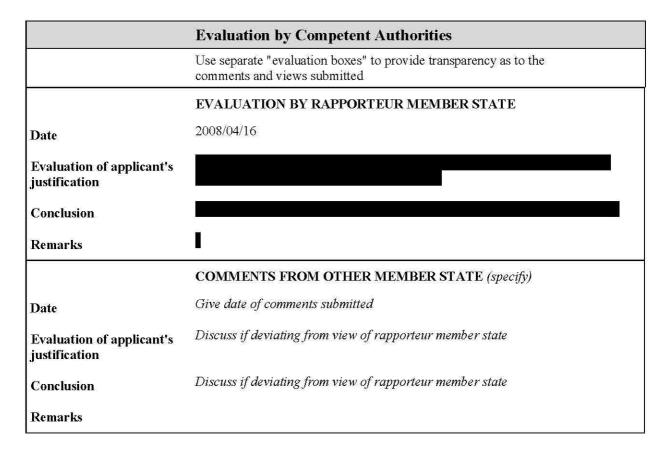
Task Force "2-Propanol" RMS: Germany	Propan-2-ol	July 2007
Section A4.2/02	Analytical Methods for Detection and Identification	
Annex Point IIA4.2	Additional information	
	ADDITIONAL INFORMATION	Official use only
Results		
		Í
References	WHO (1990) Environmental Health Criteria 103, 2-Propanol. IPCS World Health Organization 1990.	5
	Evaluation by Competent Authorities	
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted	
	EVALUATION BY RAPPORTEUR MEMBER STATE	
Date	2008/12/01	
Evaluation of results		
Conclusion		
Remarks		
	COMMENTS FROM OTHER MEMBER STATE (specify)	
Date	Give date of comments submitted	
Evaluation of results	Discuss if deviating from view of rapporteur member state	

Discuss if deviating from view of rapporteur member state

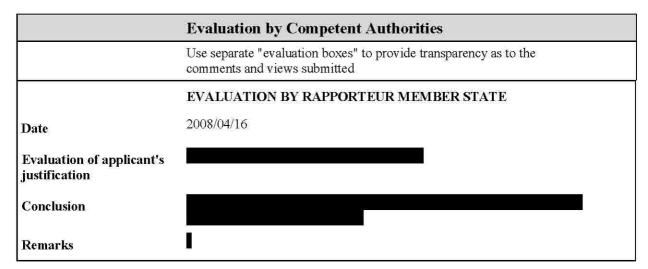
Conclusion

Remarks

Task Force "2-Propanol RMS: Germany	Propan-2-ol	July 2007
Section A4.2a Annex Point 4.2	Analytical Methods for Detection and Identification Determination of propan-2-ol in soil	
	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Official use only
Other existing data []	Technically not feasible [] Scientifically unjustified [X]	
Limited exposure []	Other justification []	
Detailed justification:		
References:		



Task Force "2-Propanol" RMS: Germany	Propan-2-ol	July 2007
Section A4.2c Annex Point 4.2	Analytical Methods for Detection and Identification Determination of propan-2-ol in water	
	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Official use only
Other existing data []	Technically not feasible [] Scientifically unjustified [X]	
Limited exposure []	Other justification []	
Detailed justification:		
References:		



Task Force "2-Propanol" RMS: Germany	Propan-2-ol	July 2007
	COMMENTS FROM OTHER MEMBER STATE (specify)	
Date	Give date of comments submitted	
Evaluation of applicant's justification	Discuss if deviating from view of rapporteur member state	
Conclusion	Discuss if deviating from view of rapporteur member state	
Remarks		

Task Force "2-Propanol" RMS: Germany	Propan-2-ol	July 2007
Section A4.2d	Analytical Methods for Detection and Identification	
Annex Point 4.2	Determination of propan-2-ol in animal and human body fluids and tissues	
	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Official use only
Other existing data [X]	Technically not feasible [] Scientifically unjustified [X]	
Limited exposure []	Other justification []	
Detailed justification:		
References:		

	Evaluation by Competent Authorities
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted
	EVALUATION BY RAPPORTEUR MEMBER STATE
Date	2008/04/16
Evaluation of applicant's justification	
Conclusion	
Remarks	
	COMMENTS FROM OTHER MEMBER STATE (specify)
Date	Give date of comments submitted
Evaluation of applicant's justification	Discuss if deviating from view of rapporteur member state
Conclusion	Discuss if deviating from view of rapporteur member state
Remarks	

Task Force "2-Propanol RMS: Germany	Propan-2-ol	July 2007
Section A4.3 Annex Point 4.3	Analytical methods including recovery rates and the limits of determination for the active substance, and for residues thereof, in/on food or feedstuffs and other products where relevant	·
	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Official use only
Other existing data []	Technically not feasible [] Scientifically unjustified [X]	
Limited exposure [X]	Other justification []	
Detailed justification:		
References:		

	Evaluation by Competent Authorities
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted
	EVALUATION BY RAPPORTEUR MEMBER STATE
Date	2008/04/16
Evaluation of applicant's justification	
Conclusion	
Remarks	ī
	COMMENTS FROM OTHER MEMBER STATE (specify)
Date	Give date of comments submitted
Evaluation of applicant's justification	Discuss if deviating from view of rapporteur member state
Conclusion	Discuss if deviating from view of rapporteur member state
Remarks	

leakage, enters the cytoplasm and destroys the inner structure of the cell molecules and of the cytoplasm's proteins. This process (referred to as denaturation) and the enzymes' coagulation leads to a loss of cellular/virus activity resulting in the cell's or virus' inactivation.

conditions, , e.g. presence of organic material at the application site.

5.4.2 Time delayRapid inactivation of the target micro-organisms/viruses without time delay due to the unspecific mode of action (topical disinfectant); time required for sufficient inactivation strongly dependent on the formulation, concentrations of propan-2-ol contained in the applied biocidal product, the type of target organisms and on the specific use

Section A5

Effectiveness against target organisms and intended uses

5.5 Field of use envisaged (IIA5.5)

MG01: Disinfectants and general biocidal products

Product type PT01: Human hygiene biocidal products

Product type PT02: Private area and public health area disinfectants

and other biocidal products

Product type PT04: Food and feed area disinfectants

5.6 User (IIA5.6)

Industrial

Propan-2-ol is a common solvent used e.g. in the preparation of medical products and pharmaceuticals (eg, anitseptics; disinfectants; rubbing alcohol; anesthetics, iodine tinctures, bathing solutions for surgical sutures and dressings), oils, gums, waxes, resins, alkaloids, cements, primers, varnishes, paints, printing inks, cosmetics and various aerosol formulations (e.g. hair sprays, floor detergents, show polishes, insecticides, window cleaners, waxes and polishes, paints, automotive products, insect repellents, flea and tick spray, air refreshers, disinfectants) and beverages.

Professional

Propan-2-ol is used by professionals as a broad-spectrum microbicide in numerous areas.

General public

Non-professional use as a broad-spectrum microbicide in numerous areas (e.g. surface and skin disinfection).

5.7 Information on the occurrence or possible occurrence of the development of resistance and appropriate management strategies

As propan-2-ol is not specific for just one target the development of sensu strictu true resistances is not to be expected (see IIIB5, 5.11).

5.7.1 Development of resistance

(IIA5.7)

True resistances are not reported.

5.7.2 Management strategies

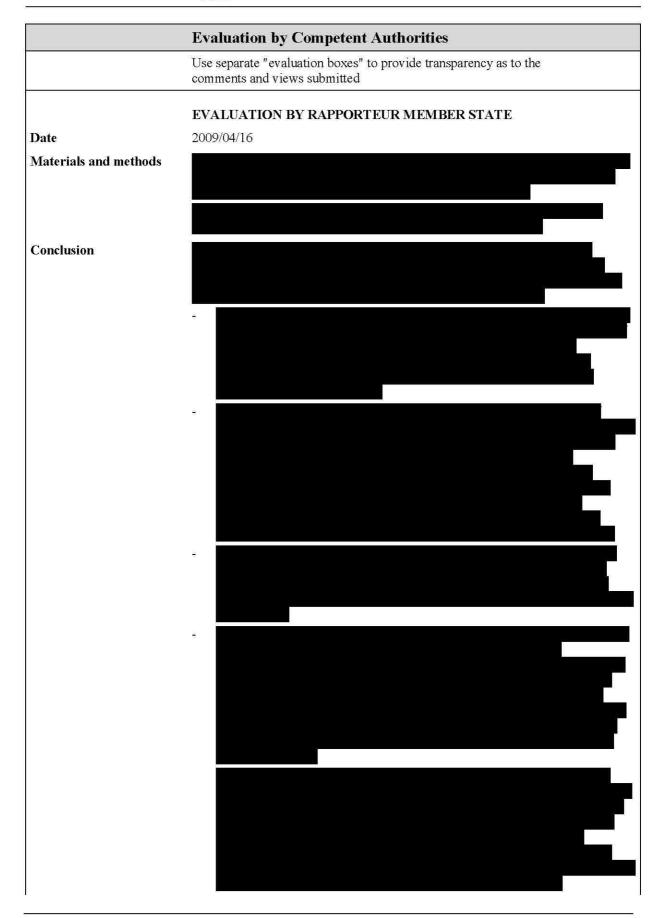
n.a.

5.8 Likely tonnage to be placed on the market per year (IIA5.8)

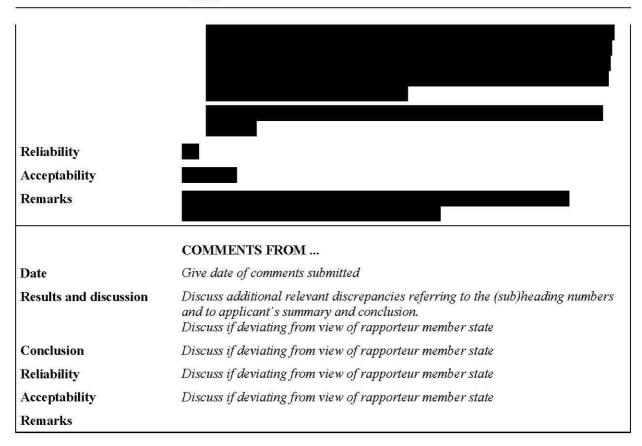


Section A5 Effectiveness against target organisms and intended uses

Propan-2-ol



Section A5 Effectiveness against target organisms and intended uses



Task Force "2-Propanol"	Propan-2-ol	July 2007
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Section 5.3: Summary table of experimental data on the effectiveness of the active substance against target organisms at different fields of use envisaged, where applicable

Function	Field of use envisaged	Test substance	Test organism(s)	Test method	Test conditions	Test results: effects, mode of action, resistance	Reference*)
		Aqueous solution of propan-2-ol	Pseudomonas aeruginosa Staphylococcus aureus Enterococcus hirae Escherichia coli Representative for Gram negative and Gram positive bacteria.	The bactericidal efficacy of propan-2-ol was characterized following a European standard guideline (EN1276).	Quantitative suspension test employing vegetative bacterial cells in the presence of organic load.		
		Aqueous solution of propan-2-ol	Staphylococcus aureus Enterococcus faecium Proteus mirabilis Pseudomonas aeruginosa Mycobacterium terrae Candida albicans Aspergillus niger Representative for Gram positive, Gram negative bacteria and yeasts and moulds.	The microbicidal efficacy of propan-2-ol was characterized following an established European test methods for the antimicrobial activity of disinfectants in food hygiene, 1987).	Quantitative suspension test employing vegetative bacterial and yeast cells and mould conidia in the presence of organic load.		
		Aqueous solution of propan-2-ol	Candida albicans Aspergillus niger Representative for fungi.	The fungicidal efficacy of propan-2- ol was characterized following an established guideline	Employed was a quantitative suspension test in the presence of organic load.	J.	

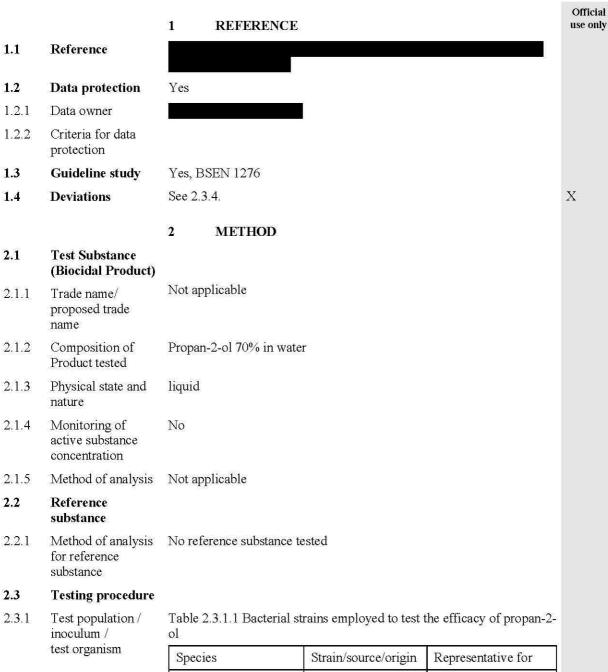
Task Force "2-Propanol"	Propan-2-ol	July 2007
RMS: Germany	- 100 Paris - 100	

		method (EN1650).		
Aqueous solution of propan-2-ol	Herpes simplex virus Representative for enveloped viruses.	The virucidal efficacy of propan-2-ol was tested using a well established laboratory method.	Employed was a carrier test method simulating practical conditions by immobilizing the test virus on a glass surface.	
Aqueous solution of propan-2-ol	Feline calicivirus Representative for non-enveloped viruses.	The virucidal efficacy of propan-2-ol was tested using an established guideline method.	Employed was a suspension test.	

Section A5.3/01 Annex Point IIA5.3

Efficacy Data

Gram-positive and gram-negative bacterial cells in the presence of organic load



Species	Strain/source/origin	Representative for
Pseudomonas aeruginosa	ATCC 15442	gram-negative bacteria
Staphylococcus aureus	ATCC 6538	gram-positive bacteria
Enterococcus hirae	ATCC 8043	gram-positive bacteria
Escherichia coli	ATCC 10536	gram-negative bacteria

The suspension contained 1.65 x 10E8 - 3.55 x 10E8 CFUs/ml

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	on A5.3/01 x Point IIA5.3	Efficacy Data Gram-positive and gram-negative bacterial cells in the presence of organic load	
2.3.2	Test system	Quantitative suspension test under conditions representative of practica use (e.g. CEN - Phase 2, Step1)	Í
2.3.3	Application of TS	As prescribed by guideline	
2.3.4	Test conditions	As prescribed by guideline, 3 different concentrations tested (50, 75% and 100% dilution of 70% propan-2-ol), Glass distilled water was used for dilution instead of sterile hard water; bovine serum albumin at 3g/l served as organic load; test was run at 20°C; as neutralizer to stop the effect of the biocide a solution containing 3 g/l lecithin; 30 g/l polysorbat 80; 5 g/l sodium thiosulfate, 1g/l L-histidin, 30 g/l saponin it tryptone sodium chloride solution was employed	
2.3.5	Duration of the test / Exposure time	5 min	
2.3.6	Number of replicates performed	As prescribed by guideline	
2.3.7	Controls	As prescribed by guideline	
2.4	Examination		
2.4.1	Effect investigated	Reduction in viability of respective test organism using a quantitative suspension test (phase 2/step1) as prescribed by the guideline EN 1276	
2.4.2	Method for recording / scoring of the effect	Determining the number of CFUs of respective test organism in test suspension before and after exposure to the test substance	
2.4.3	Intervals of examination	CFUs determined once after termination of exposure	
2.4.4	Statistics	As prescribed by guideline	
2.4.5	Post monitoring of the test organism	No	
		3 RESULTS	
3.1	Efficacy	In accordance with the guideline EN 1276 the product possesses bactericidal activity against the tested organisms	
3.1.1	Dose/Efficacy curve	Not applicable	
3.1.2	Begin and duration of effects		
3.1.3	Observed effects in the post monitoring phase	Not applicable	
3.2	Effects against organisms or objects to be protected	None reported	
3.3	Other effects	None reported.	
3.4	Efficay of the reference substance	Not applicable	

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Section A5.3/01 Annex Point IIA5.3

Efficacy Data

Gram-positive and gram-negative bacterial cells in the presence of organic load

3.5 Tabular and/or graphical presentation of the summarised results

Table 3.5.1 Reduction of viable cells after exposure to aqueous propan-2-ol solution

Species/strain	Concentration of test product	Viability reduction (cfu/ml)
Pseudomonas aeruginosa	50%	<10E5
	75%	>10E5
	100%	>10E5
Staphylococcus aureus	50%	<10E5
	75%	>10E5
	100%	>10E5
Enterococcus hirae	50%	>10E5
	75%	>10E5
	100%	>10E5
Escherichia coli	50%	<10E5
	75%	>10E5
	100%	>10E5

3.6 Efficacy limiting factors

3.6.1 Occurrences of resistances

None reported

3.6.2 Other limiting factors

None reported

4 RELEVANCE OF THE RESULTS COMPARED TO FIELD CONDITIONS

4.1 Reasons for laboratory testing

The bactericidal activity of the product was tested using two grampositive - *Staphylococcus aureus* and *Enterococcus hirea* – and two gram-negative bacterial species - *Pseudomonas aeruginosa* and *Escherichia coli* in accordance with a standard guideline method. The data obtained in this study are relevant for the intended field of use.

4.2 Intended actual scale of biocide application

Not stated

- 4.3 Relevance compared to field conditions
- conditions
- 4.3.1 Application method

The test conditions of the quantitative suspension test (phase 2/step1) in the presence of organic load as proposed by CEN are representative for

the actual conditions during practical use of the product.

4.3.2 Test organism

The test organisms used in this study representing both gram-positive and gram-negative bacterial species are appropriate representatives for the target organisms in the intended field of use.

Section A5.3/01 Annex Point IIA5.3

Efficacy Data

Gram-positive and gram-negative bacterial cells in the presence of organic load

4.3.3 Observed effect

The obtained efficacy result of the test product in this study using 2 gram-positive and 2 gram-negative bacterial species under simulated use conditions in the presence of organic load is important for evaluating the bactericidal activity of the product in the intended field of use.

4.4 Relevance for read-across

5 APPLICANT'S SUMMARY AND CONCLUSION

5.1 Materials and methods

The bactericidal activity of propan-2-ol 70% in water was evaluated using a generally accepted quantitative suspension test (phase 2/step1) as prescribed by the guideline EN 1276. Two gram positive (S. aureus, E.hirea) as well as two gram negative bacterial species (P.aeruginosa, E.coli) were used as test organisms. The suspension test was carried out in the presence of organic load (3g/l bovine serum albumin) to simulate practical conditions as prescribed by the guideline. The test was carried out at 20°C for an exposure time of 5min at various concentrations (50%, 75% and 100%). The reduction in viability was determined via CFU count.

5.2 Reliability

5.3

Assessment of efficacy, data analysis and interpretation The results of this study show that 70% propan-2-ol in water tested in the presence of organic load (3g/l bovine albumin) and at an exposure duration of 5min was effective against the bacterial species tested. However, the study showed that the product was not effective against Staphylococcus aureus, E. coli and Pseudomonas aeruginosa at a concentration of 50%. In this case the criteria for efficacy as required by the guideline were not fulfilled. Propan-2-ol concentrations of 75%-100% were effective against all tested organisms used in the study and the criteria for efficacy were fulfilled.

X

5.4 Conclusion



5.5 **Proposed efficacy** specification

	Evaluation by Competent Authorities
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted
	EVALUATION BY RAPPORTEUR MEMBER STATE
Date	2008/09/23
Materials and methods	
Conclusion	
Reliability	
Acceptability	
Remarks	
	COMMENTS FROM
Date	Give date of comments submitted
Results and discussion	Discuss additional relevant discrepancies referring to the (sub)heading numbers and to applicant's summary and conclusion. Discuss if deviating from view of rapporteur member state
Conclusion	Discuss if deviating from view of rapporteur member state
Reliability	Discuss if deviating from view of rapporteur member state
Acceptability	Discuss if deviating from view of rapporteur member state
Remarks	

Appendix 1: CA-Tables



Task Force "2-Propanol" Propan-2-ol **July 2007 RMS: Germany**

Section A5.3/02 **Annex Point IIA5.3** **Efficacy Data**

Gram-positive and Gram-negative bacterial cells and fungi in the presence of organic load

> Official REFERENCE use only

1.1 Reference

Vergelijkend onderzoek naar de desinfekterende werking van alcoholen in de Europese suspensie test.

1.2 Data protection

Not stated

1

1.2.1 Data owner

1.2.2 Criteria for data protection

Not stated

Guideline 1.3 study

Yes, Test methods for the antimicrobial activity of disinfectants in food hygiene, European Council, 1987.

1.4 **Deviations** See 2.3.5.

2 **METHOD**

2.1 Test Substance (Biocidal Product)

2.1.1 Trade name/ Not applicable

proposed trade

name

2.1.2 Composition of Product tested

Propan-2-ol (p.a, Merck 9634) at different concentrations in water

2.1.3 Physical state and nature

liquid

No

2.1.4 Monitoring of

active substance concentration

2.1.5 Method of analysis Not applicable

> 2.2 Reference substance

Propan-1-ol and ethanol tested in parallel.

2.2.1 Method of analysis no for reference

substance

2.3 **Testing** procedure

2.3.1 Test population / inoculum / test organism

Table 2.3.1.1 Bacterial strains employed to test the biocidal efficacy of propan-2-ol

relance or		
Species/strain	Source/origin	Representative for
Pseudomonas aeruginosa	ATCC 15442	Gram negative bacteria
Staphylococcus aureus	ATCC 6538	Gram positive bacteria
Enterococcus faecium	DVG 8582	Gram positive

Section A5.3/02 **Annex Point IIA5.3**

Efficacy Data

Gram-positive and Gram-negative bacterial cells and fungi in the presence of organic load

		bacteria
Proteus mirabilis	ATCC 14153	Gram negative bacteria
Mycobacterium terrae	ATCC 15755	Gram positive bacteria
Candida albicans	ATTC 10231	Yeasts
Aspergillus niger	ATCC 16404	Moulds

The bacterial suspensions contained approx. 10E8 CFUs/ml, the yeast suspension approx. 10E7 CFUs/ml, the conidial suspension approx. 10E7 CFUs/ml. Stock cultures of all strains but A. niger and M. terrae were kept on tryptone soy agar. M. terrae was kept on Middlebrook 7H10 Agar with 10% OADC whilst A. niger was kept on malt extract agar. Working cultures (2 subsequent times 24h growth on TSA at 32°C) were used to prepare suspensions for all bacterial strains (exception M. terrae) and the yeast by using glass beads and glass wool filtration. M. terrae suspensions were obtained from 7d stock cultures using glass beads and subsequent filtration with glass wool. A. niger conidia were harvested from 4d stock cultures using 0.6% Tergitol 7, harvested by centrifugation (20 min @ 2000 g). All suspensions were

- prepared in saline with 0.1% peptone. 2.3.2 Quantitative suspension test under conditions representative of practical Test system use (e.g. CEN - Phase 2, Step1) As prescribed by guideline, diluted in water of standard hardness. 2.3.3 Application of TS 2.3.4 Test conditions Concentrations tested (20 up to 80% propan-2-ol (v/v)), dilution in sterile hard water; bovine serum albumin at 0.03% served as organic load; test was run at 20°C+/-1°C; dilution in neutralizer solution used to stop the effect of the biocide. 2.3.5 Duration of the test 2 and 5 min / Exposure time As prescribed by guideline 2.3.6 Number of replicates performed 2.3.7 Controls As prescribed by guideline 2.4 Examinati on 2.4.1 Effect investigated Reduction in viability of respective test organism using a quantitative
- suspension test (phase 2/step1) as prescribed by the guideline employed.
- Determining the number of CFUs of respective test organism in test 2.4.2 Method for recording / scoring suspension before and after exposure to the test substance of the effect
- Intervals of 2.4.3 CFUs determined once after termination of exposure examination
- Statistics 2.4.4
- 2.4.5 Post monitoring of the test organism

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July 2007

Section A5.3/02 **Annex Point IIA5.3**

Efficacy Data

Gram-positive and Gram-negative bacterial cells and fungi in the presence of organic load

3 RESULTS

3.1 Efficacy Propan-2-ol exhibited biocidal activity for all organisms tested.

Dose/Efficacy 3.1.1 curve

Not applicable

3.1.2 Begin and duration

of effects

Observed effects in Not applicable 3.1.3 the post monitoring

3.2

phase

Effects against organisms or objects to be protected

None reported

3.3 Other effects None reported.

3.4 the reference

substance

Efficay of Propan-1-ol was more effective than propan-2-ol which was more effective than ethanol (exception A. niger).

3.5 Tabular and/or graphical presentati on of the summaris ed results

Table 3.5.1 Reduction of CFUs/ml after exposure to aqueous propan-2ol solution

Species/strain	Exposure time (min)	Concentration of test product (%, v/v)	Viability reduction (log RF CFUs/ml)
Pseudomonas aeruginosa	2	20	1.8
		30	>=5
	5	20	2.5
		30	>=5
Staphylococcus aurenus	2	20	<=0.2
		30	>=5
		40	>=5
	5	20	0.3
		30	>=5
		40	>=5
Enterococcus faecium	2	20	<=0.2
		30	>=5
		40	>=5
	5	20	<=0.2
		30	>=5

Section A5.3/02 Annex Point IIA5.3

Efficacy Data

Gram-positive and Gram-negative bacterial cells and fungi in the presence of organic load

		-	
		40	>=5
Proteus mirabilis	2	20	2.5
		30	>=5
	5	20	2.9
		30	>=5
Mycobacterium terrae	2	30	2.9
		40	>=5
		50	>=5
	5	30	>=5
		40	>=5
		50	>=5
Candida	2	20	<=0.2
albicans		30	>=5
		40	>=5
	5	20	<=0.2
		30	>=5
		40	>=5
Aspergillus niger	2	40	<=0.2
		50	<=0.2
		60	0.6
		70	1.2
		80	2.1
	5	40	<=0.2
		50	0.5
		60	1
		70	1.8
		80	3.2

3.6 Efficacy limiting factors

3.6.1 Occurrences of resistances

None reported

3.6.2 Other limiting factors

None reported

4 RELEVANCE OF THE RESULTS COMPARED TO FIELD CONDITIONS

4.1 Reasons for

The microbicidal activity of the product was tested using three Gram positive (Staphylococcus aureus, Mycobacterium terrae and

Section A5.3/02 Annex Point IIA5.3

Efficacy Data

Gram-positive and Gram-negative bacterial cells and fungi in the presence of organic load

laborator y testing

Enterococcus faecium) and two Gram negative bacterial species (Pseudomonas aeruginosa and Proteus mirabilis) as well as two fungal species (Candida albicans and Aspergillus niger). The data obtained in this study are relevant for the intended field of use.

4.2 Intended actual

d Not stated

- actual scale of biocide applicatio n
- 4.3 Relevance compared to field conditions
- 4.3.1 Application method

The test conditions of the quantitative suspension test (phase 2/step1) in the presence of organic load are representative for the actual conditions during practical use of the product.

4.3.2 Test organism

The test organisms used in this study representing both gram-positive and gram-negative bacterial as well as fungal species are appropriate representatives for the target organisms in the intended field of use.

4.3.3 Observed effect

The obtained efficacy result of the test product in this study using 5 different bacterial and 2 fungal species under simulated use conditions in the presence of organic load is important for evaluating the bactericidal activity of the product in the intended field of use.

4.4 Relevance for readacross

5 APPLICANT'S SUMMARY AND CONCLUSION

5.1 Materials and methods

The bactericidal activity of propan-2-ol in water was evaluated using a generally accepted suspension test (phase 2/step1). Two gram positive (S. aureus, E. faecium, M. terrae), two gram negative bacterial species (P. aeruginosa, P. mirabilis) and two fungi (A. niger, C. albicans) were used as test organisms. The suspension test was carried out in the presence of organic load (0.03% bovine serum albumin) to simulate practical conditions. The test was carried out at 20°C for an exposure time of 2 and 5min at various concentrations (20 - 80%). The reduction in viability was determined per CFU count.

- 5.2 Reliability
- 5.3 Assessme nt of efficacy, data analysis and interpreta tion

The results of this study show that 30% propan-2-ol in water tested in the presence of organic load (0.03% bovine serum albumin) and at an exposure time of 5min was effective against the bacterial and fungal species tested in the study. However, sufficient effectivity against *Aspergillus niger* conidia required test substance concentrations of >= 80%.

5.4 Conclusio n

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